Screening of Phytochemical Secondary Metabolites of *Muntingia Calabura*: a Potential as Hepatoprotector

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Abstract

Muntingia calabura is one of the plants employed to produce herbalbased treatments. Muntingia calabura leaves are traditionally used as an alternative medicine due to their secondary metabolites. The maceration method extracted Muntingia calabura leaves using 96% ethanol solvent for 3 x 24 hours. The fractionation process was carried out using a separating funnel method with different polarities, such as n-hexane, ethyl acetate, and ethanol-water. Thinlayer chromatography (TLC) was used to confirm the phytochemical screening. TLC conditions under UV light 254 and 366 nm using solvents, such as chloroform: methanol (alkaloids), butanol: acetic acid: water (flavonoids), chloroform:methanol: water (saponins), and chloroform: methanol (phenolic). The phytochemical screening results of extracts and Muntingia calabura fractions contained secondary metabolites, such as alkaloids, flavonoids, tannins, saponins, and phenolics. TLC results showed that n-hexane fraction contained flavonoid and saponin compounds; ethyl acetate fraction contained alkaloids, flavonoids, saponins and phenolic compounds; and ethanol-water fraction contained alkaloids, flavonoids, saponins, and phenolics. Muntingia calabura leaves indicated the potential as herbal medicine by containing secondary metabolites.

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INTRODUCTION

Natural resources have benefits for various purposes, such as developing herbalbased medicines¹. The utilization of natural ingredients for traditional medicine has been consumed by a large number of people in the world for many years². In addition, traditional medicine is also used as a health supplement by 80% of the population in developing countries around the world, and about 85% of ingredients in traditional medicine involve plant extracts³. One of the plants used as traditional medicine is *Muntingia calabura*.

Muntingia calabura belongs to the *Elaeocarpaceae* family, widely used in traditional medicine. People in Indonesia usually eat the fruit directly due to its sweet taste⁴. Meanwhile, the Peruvian people have used the flowers and stem as an antiseptic to reduce swelling of the prostate gland and reduce headaches and fever⁵. Several studies have shown that extracts of *Muntingia calabura* had a

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variety of activities that had been tested, such as antioxidant⁶, anticancer⁷, antidiabetic⁸, antibacterial⁹, and antiinflammatory¹⁰.

Based on this information, this study aims to determine the secondary metabolites contained in the *Muntingia calabura* leaf fraction so that further tests can be carried out both in vivo and in vitro as hepatoprotection.

METHOD

This study is a laboratory experiment aiming to discover the secondary metabolites found in *Muntingia calabura* plants. This study was conducted at the Pelita Mas Palu School of Pharmacy's Phytochemical Laboratory.

Research Tools and Materials

The tools used in this study were an oven (Memmert Universal), rotary evaporator (Ryela N-100), water bath (Eyela SB-1000), TLC plate, Photometer 5010 (Riele-Germany), UV lamp, and glassware (pyrex).

The materials used in this study were *Muntingia calabura* leave, ammonia (Merck), aquadest, anhydrous acetic acid (Merck), HCl (Merck), concentrated HCl (Merck), H2SO4 (Merck), butanol (Merck), ethanol 96% (Merck), ethyl acetate (Bratachem), FeCl₃ (Merck), chloroform (Merck), methanol (Merck), NaCl (Merck), n-hexane (Merck), ragendroff Merck), magnesium powder P (Merck).

Determination

The identification of *Muntingia calabura* was carried out at UPT. Biological Resources Tadulako University, Central Sulawesi.

Preparation of *Muntingia calabura* Extracts and Fractions

The maceration method extracted *Muntingia calabura* Simplicia using 96 percent ethanol solvent for 3×24 hours. To obtain a thick *Muntingia calabura* extract, the maceration results were filtered and concentrated using a rotary evaporator at a temperature of $40-50^{\circ}C^{11}$.

The thick ethanol extract of Muuntingia calabura leaves was fractionated using non-polar, semi-polar and polar solvents (n-hexane, ethyl acetate and water). First, the thick Muntingia calabura extract was fractionated in a separating funnel with nhexane and water (1:3) and thoroughly shaken. After that, it was left until two layers, the n-Hexane layer and the water layer had formed. To get the n-hexane fraction, the process was repeated three times. The water layer was fractionated three times with ethyl acetate (3:1), yielding the water and ethyl acetate fractions. To obtain a thick fraction of Muntingia calabura leaves, the results of the n-hexane and ethyl acetate fractions were condensed with a rotary evaporator at a temperature of 40-50°C¹¹.

Phytochemical Screening of *Muntingia* calabura Extract and Fraction

Alkaloids Test

o.5 grams of the extract was weighed, and the thick fraction of *Muntingia calabura* leaves were weighed and placed in an Erlenmeyer with 5 ml chloroform and 5 ml ammonia. It was then heated on a water bath, shaken and filtered. 5 drops of 2N H2SO4 were added to the filtrate in a test tube. Dragendorff's reagent was added to each tube containing the filtrate, along with several mL of HCl, and the tubes were filtered. Dragendorff's reagent produced red-orange precipitation, which indicated a positive reaction¹².

Flavonoids Test

o.5 grams of the extract was weighed, and the thick fraction of *Muntingia calabura* leaves was added with 10 ml of distilled water and heated for 1 minute on a water bath. It was then filtered and dissolved in 1 ml of ethanol (95%) with magnesium P powder. If a crimson hue appeared after being dissolved in 10 mL of strong hydrochloric acid, it would indicate the presence of flavonoids¹².

Saponins Test

o.5 grams of the extract was weighed, and a fraction of *Muntingia calabura* leaves were put into a test tube. After that, the 10 mL of hot water was added, let cool, and was shaken vigorously for 10 seconds. Saponins would be present if a foam formed and lasted for at least 1 minute at the height of 10 cm or if 1 drop of 2N hydrochloric acid did not evaporate after being added¹².

Tannins Test

o.5 grams of extract and a thick fraction of *Muntingia calabura* leaves were weighed and placed in a test tube with 20 mL hot water and 3 drops of a 10% NaCl solution. If a blue-black color appeared after adding a FeCl₃ solution, it would mean the presence of tannin¹².

Phenolic Test

o.5 grams of extract and a thick fraction of *Muntingia calabura* leaves were weighed and dissolved in 5 mL of water. If a dark green color shift occurred after adding a few drops of 5% FeCl₃ solution, it would mean the existence of phenolic¹².

Thin-Layer Chromatography of Qualitative Test

Alkaloids Test

In the development of chloroform: methanol (85:15), the sample was

detected on a silica gel plate. Detection of the presence of alkaloids was conducted under UV light 254-366 nm. Ensure that the plate was sprayed with Dragendorff's reagent, and spots would appear if it were positive for alkaloids¹³.

Flavonoids Test

In the development of butanol: acetic acid: water (4:1:5), the sample was detected on a silica gel plate. Detection of the presence of flavonoids was conducted under UV light 254-366 nm. It was then sprayed with AlCl₃ staining solution in chloroform; yellow color indicated the presence of flavonoids¹³.

Saponins Test

The sample was detected on a silica gel plate in developing chloroform: methanol: water (40:50:10). It was then sprayed with a mixture of anisaldehyde and sulfuric acid reagent, which produced a purplish-blue tint with a hint of yellow¹³.

Phenolic Test

In the development of chloroform: methanol (9:1), the sample was detected on a silica gel plate. It was then sprayed with FeCl₃ reagent and formed a blackish blue, green or turquoise color¹³.

RESULTS AND DISCUSSION

The test material in this study was *Muntingia calabura* leaves obtained from the city of Palu and was started from June to August 2020. The determination aimed to determine the accuracy of the test material employed. Determination of *Muntingia calabura* leaves was carried out at UPT. Biological Resources Tadulako University, Central Sulawesi. The analysis revealed that the *Muntingia calabura* leaves utilized in the study belonged to the genus *Elaeocarpaceae*.

Muntingia calabura leaves were made by maceration using ethanol as a solvent¹⁴.

The thick extract of *Muntingia calabura* leaves obtained from the simplicia maceration was 45 grams with a yield value of 6.4%. Furthermore, the thick extract of *Muntingia calabura* leaves was fractionated using 3 solvents, namely nhexane, ethyl acetate and ethanol-water. The weight of each fraction obtained was the n-hexane fraction of *Muntingia calabura* leaves of 20.29 grams with a yield value of 4.05%, the ethyl acetate fraction of *Muntingia calabura* leaves of 13.91 grams with a yield value of 2.7% and the ethanol-water fraction of *Muntingia* *calabura* leaves of 29.17 grams with a yield value of 5.83%.

The class of secondary metabolites found in *Muntingia calabura* leaves extract as bioactive chemicals expected to have a role in delivering hepatoprotective effects was determined through phytochemical testing. The ethanol extract of *Muntingia calabura* leaves contained positive alkaloids, flavonoids, saponins, tannins, and phenolics, according to the results of phytochemical screening (Table I).

Chemical Compounds	Reagent	Result	Exp.
Alkaloids	Dragendorf	Red	(+)
Flavonoids	Mg powder + HCl	Red yellow	(+)
Saponins	Water + HCl	Foam ± 1 cm	(+)
Tannins	FeCl ₃ + NaCl 10%	Blackish green	(+)
Phenolic	FeCl ₃ 1%	Blackish green	(+)

Table 1. The Results of Ph	ytochemical Screenin	g of <i>Muntingia</i>	calabura Extract
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Note: (+): Detected

(-) : Not detected

As *Muntingia calabura* leaves extract contained a wide range of polar, semipolar, and non-polar secondary metabolite chemicals, the fractionation method was used to separate some of these compounds based on their polarity. Fractionation was carried out in stages, starting with separating non-polar compounds using n-hexane as a non-polar solvent, then semi-polar compounds using ethyl acetate as a semi-polar solvent, and finally using water as a polar solvent to compounds¹⁵. attract polar The phytochemical screening of the obtained revealed that fractions alkaloids. flavonoids, saponins, tannins, and polyphenols were present in the Muntingia calabura fractions (Table 2).

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		Result			
Chemical Compound	Reagent	n-Hexane	Ethyl Acetate	Ethanol-Water	
	_	Fraction	Fraction	Fraction	
Alkaloids	Dragendorf	-	+	+	
Flavonoids	Mg and HCl	+	+	+	
Saponins	Water and HCl	+	+	+	
Tannins	FeCl ₃ + NaCl	-	+	+	
Phenolic	FeCl ₃	-	+	+	

Note: (+): Detected

(-) : Not detected

Thin-layer chromatography (TLC) was performed to observe the chromatogram

pattern of the *Muntingia calabura* fraction. TLC can be identified by separating the Muntingia calabura fraction and the eluent. The eluents used in determining the chromatogram of the Muntingia calabura fraction were alkaloid compounds using a mixture of chloroform: methanol (85:15), flavonoid compounds using a mixture of butanol: acetic acid: water (4: 1: 5), saponin compounds using a mixture of chloroform: methanol: water (40:50:10) and phenolic compounds using a mixture of chloroform: ethanol (9:1). The observations were conducted under UV light 254 and 366 nm. At UV 254 nm, the plate fluoresced while the sample appeared dark in color. Meanwhile, the stain fluoresced at UV 366 nm, and the plate appeared dark in color. It was then followed by spraying using 10% H2SO4 in methanol. The results of TLC showed that the *Muntingia calabura* fraction contained alkaloids, flavonoids, saponins and phenols, which were indicated by purple spots (Table 3 and figure 1).

Chemical Compound	Eluent	n-Hexane Fraction	Ethyl Acetate Fraction	Ethanol-Water Fraction
Alkaloids	Chloroform: Methanol (85: 15)	-	+	-
Flavonoids	Butanol:Acetic acid:Water (4 :1 : 5)	+	+	+
Saponins	Chloroform:Methanol:Water (40: 50: 10)	+	+	+
Phenolic	Chloroform: Methanol (9: 1)	-	+	+

Table 3. The Results of Thin-Layer Chromatography

(-) : Not detected





Figure 1. Thin-layer chromatography test using UV light with a wavelength of 254 nm and 366 nm (a) Alkaloids; (b) Flavonoids; (c) Saponins; (d) Phenolic

The result of TLC profiling is summarized in figure 1. N-hexane fraction Muntingia calabura showed the presence of flavonoid (Rf o.8) and saponin (Rf o.8). Ethyl acetate fraction Muntingia calabura showed the presence of alkaloid (Rf o.7), flavonoid (Rf o.6), saponin (Rf o.8) and phenolic (Rf o.8). Ethanol-water fraction Muntingia calabura showed the presence of flavonoid (Rf 0.5), saponin (Rf 0.7), and phenolic (Rf 0.3). TLC is usually done for better identification of the bioactive compounds. Different Rf values of the compounds provide an idea about their polartity that may also help select a particular solvent system to isolate any compound from the plant fraction further using chromatographic and spectroscopic. Compounds with a high Rf value in a less polar solvent system have low polarity while those with a low Rf value have high.

These findings suggested that Muntingia calabura fractions could be evaluated in vivo and in vitro for hepatoprotective properties. Research conducted by (Haki, 2019) revealed that the Muntingia calabura extract could reduce the ALT enzyme of mice although it had not reached the normal value¹⁶. Furthermore, (Hakim, 2012) research stated that the Muntingia calabura extract dose of 420 mg/200 gram and 84 mg/200 gram could prevent increased levels of the enzyme ALT in acetaminophen-induced rats¹⁷. This information is critical for a simple standardization in evaluating activity as herbal-based medicines.

CONCLUSION

Based on the research results, it is confirmed that *Muntingia calabura* contained alkaloids, tannins, saponins, and phenolics, as evidenced by TLC results. As a result, *Muntingia calabura* played hepatoprotective herbal components that could be investigated in vivo and in vitro.

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