

Synthesis, Characterization and Investigation the Antibacterial Activity of New Heterocyclic Compounds Derived From 4-(4`-methoxy benzoyloxy) benzaldehydethiosemicarbazone

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Abstract

4-Thiazolidinone were synthesized by three steps, the reaction of ansoyl chloride with 4hydroxy benzaldehyde to give 4-(4'-methoxy benzoyloxy) benzaldehyde[I]. The reaction of later compound with thiosemicarbazideled to formation thiosemicarbazon [II] and the reacted thiosemicarbazone with chloro acetic acid in CH₃CO₂Na medium to yield 4- thiazelidinone compound[III]. The 4-thiazolidinone [III] was used as a key intermediates to synthesis new compounds, compound[IV] synthesized from the reaction [III] with acetic anhydride, while the reaction of compound [III] with amines to yield azo compound[V]_{a,b,c}. The azo compound reacted with benzoyl chloride or anisole chloride in basic medium to get a new esters compound[VI]_{a,b}. Also, synthesis oxazepine compound[VII],[VIII] and [IX]_{a,b,c} from heating schiff bases [III],[IV][V]_{a,b,c} with phthalic anhydride in dry benzene in cyclo-addition reaction. The compounds have been screened for their six species of bacteria were used in this study as tested organisms. These are *pseudomonas aeruginosa, klebseilla, providencea, Serratiamarcescens*(Gram negative) and *staphylococcus epidermidis, Bacillus*(Gram positive).

Keywords: thiazolidinone, Schiff bass, azo compound, 1,3-oxazepine and antibacterial activity.

Introduction

1,3-Thiazolidin-4-one is five heterocyclic system that has two heteroatoms sulfur and nitrogen at position 1,3 respectively, beside a carbonyl group at position4 [1].Thiazolidin-4-ones are usually synthesized from thiourea, thiosemicarbazides or azomethinegroup[2]. Thiazolidine ring is of considerable interest as it is a structure in various synthetic pharmaceuticals have a broad spectrum of biological activities [3-5]. Oxazepine is unsaturated non-homologous seven membered heterocyclic (containing oxygen at position 1 and nitrogen at position 3 in addition to the five carbon atoms. It is prepared by the pericyclic cycloaddition reaction of schiff bases with a suitable cyclic anhydride carboxylic acid) [6,7]. Oxazepine derivatives was found to exhibit a good variety of biological activities [8]like, anticancer [9], antifungal [10], antibacterial [11] and anticorrosion [12].Azo linkage is considered biological active moiety [13],The most important method for preparing azo compounds via the coupling reaction between diazonium salts and phenols[14].

The aim of this work is synthesis of some new oxazepine andthiazolidin4-ones derivatives containing the biologically active azo group, as attempt for increasing the biological activity and its variety. [15]

Experimental

Chemicals: All chemicals were supplied by fluka, GCC, merek and Aldrich chemicals Co. and used as received.

Techniques

FTIR spectra were recorded by using KBr disc on a Shimadzo (Ir prestige -21) ¹HNMR spectra were examined by company : Bruker , model: ultra shield 300 MHz , origin : Switzerland and are reported in DMSO as a solvent,ppm(δ), uses TMS as an internal standard were made at chemistry department , Al-Bayt University, Jordan . Hot-Stage, Gallen Kamp melting point apparatus was used for determined uncorrected melting points. **Synthesis** New compounds are synthesized according to scheme1.

Preparation of [4-(4`-methoxy benzoyloxy)benzaldehyde[I]

Ansoyl chloride (0.17g, 0.001mol) was added to a stirred solution of 4-hydroxy benzaldehyde (0.12g, 0.001mol), dry pyridine (1mL) in dimethyl formamide (DMF) (10 mL) at (5-10°C). Stirring was continued for 3hrs (at the same temperature). The resulting mixture was poured onto 20 mL of 10% hydrochloric acid. The precipitate was filtered, washed with solution of 10% NaHCO₃ and water for several times ,dried and recrystallized from ethanol.m.p76-78^oC, yield86%.

Synthesis of 4-(4 -methoxybenzoyloxy)benzaldehydethiosemicarbazone [II]

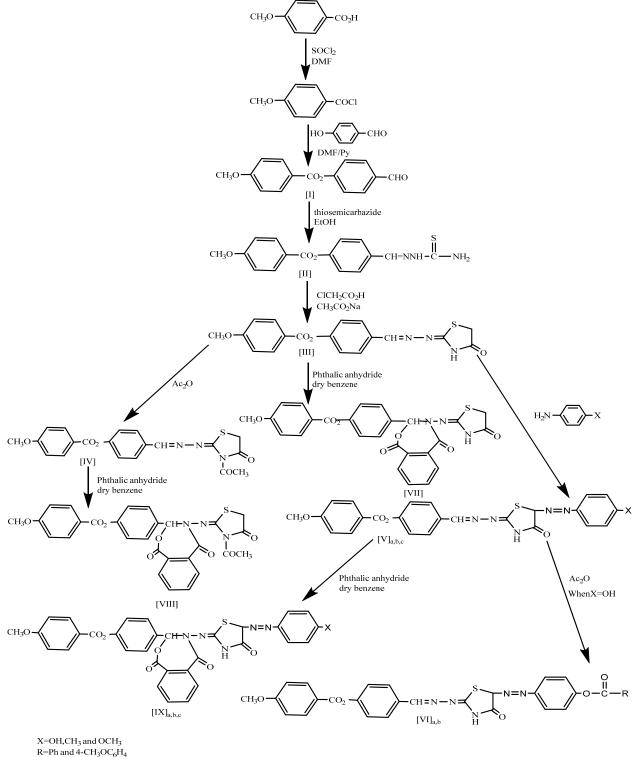
A mixture of a suitable aromatic aldehyde [I] (0.256g, 0.001mole), thiosemicarbazide (0.091g, 0.001mole) in ethanol (5mL) was heated under reflux for 4hrs, then cooled .The off white solid formed was filtered, dried and recrystallized from ethanol to give compound [III]. m.p200-202^oC, yield 97%.

Synthesis of 4-(((4-oxothiazolidin-2-ylidene)hydrazono)methyl)phenyl 4-methoxybenzoate[III]

A mixture of thiosemicarbazone[II] (0.01mol) ,chloroaceticacide (0.01mol, 0.945g) and fused sodium acetate (0.03mol,2.46g) in absolute ethanol(10 mL) was heated for 6 hrs. The mixture of reaction was poured onto (100 mL) cold water and the precipitate was filtered, washed with water (for many times), recrystallized by ethanol.

Synthesis of 4-(((3-acetyl-4-oxothiazolidin-2-ylidene)hydrazono)methyl)phenyl 4-methoxybenzoate[IV]

A solution of compound [III] (3.67g ,0.01mole) in acetic anhydride (25mL)was refluxed for 4hrs, afterward cooled and poured onto ice-water .The resulting was filtered off ,washed with water, dried and recrystallized from ethanol to yield compound [IV].



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Synthesis of azo derivatives [V]_{a, b, c}

This compound was prepared by two main steps:

A-Azofization: Into flask was placed in the ice bath (2.1g, 0.02mole) of compound amine dissolved in 12mL of concentrated hydrochloric acide (50%) and the mixture was cooled at (0-5) 0 C then added 8mL of a solution contains sodium nitrite(20%) drop by drop to the reaction, that led to formation of diazonium salt in solution.

B-Coupling :A mixture of solution (3.67g, 0.01 mole) from compound [III] in 50mL of aqueous solution NaOH (10%) was put into a 250 mL two necked round bottom flask and fitted with thermometer and a dropping funnel and placed in ice bath to be cooled $at(0-5)^{0}C$. The cold diazonium solution (step A) was added to the solution (of step B) drop by drop with stirring during two hrs. At the same temperature, we add a solution of HCl 30% to the mixture to form a precipitated compound, which was filtered, washed by cold water and recrystallized from chloroform.

Synthesis of 4-(((5-((4-(benzoyloxy)phenyl)diazenyl)-4-oxothiazolidin-2-ylidene) hydrazono)methyl)phenyl 4-methoxybenzoate and 4-((2-((4-(4-methoxybenzoyloxy) benzylidene)hydrazono)-4-oxothiazolidin-5-yl)diazenyl)phenyl 4-methoxybenzoate [VI]_{a,b}

To a stirred solution of compound $[V]_c$ (0.47g, 0.001mol), triethylamine(0.2g,0.002mol)in dried mixture of (5mL DMF:10Ml THF), was added drop wise carboxylic acid chloride(0.001mol) at(0-4) 0 C. After the addition had been completed the resulting suspension was stirred at the same temperature for 3hrs. The triethylaminehydrochloride salt was precipitate. It was filtered and the filtrate was poured with stirring into 100mL ice-water, then resulting solid was filtered and recrystallized from ethanol.

Synthesis of 1,3-oxazepine derivatives [VII],[VIII] and [IX]_{a,b,c}

A mixture of compound [III],[IV] or $[V]_{a,b,c}$ (0.48g, 0.001mole) and phthalic acid anhydrides (0.14g, 0.001mol) in dry benzene (3mL) was heating for 6 hrs. The solvent was removed and the resulting colored crystalline solid recrystallized from ethanol. The physical properties for the synthesized compounds are given in Table1.

Determination of antibacterial activity

Antibacterial activity were detected by agar well diffusion method [16]. Fresh bacterial cultures suspension equivalent of 0.5 tube McFarland turbidity standards (10^8 cfu/ml) were spread on Muller-Hinton agar plates using sterile cotton swabs. Wells of 6mm diameter were cut in solidified agar and filled with 50µl 0f each concentration. The dimethyl sulfoxide was also used as control. The plates were incubated aerobically at 37 ^o C for 24 hours, then inhibition zones diameter (mm) around wells were measured by rule. All testes were applied as doublicate.

Results and Discussion

4-(4'-methoxybenzoyloxy) benzaldehyde [I] was prepared from reaction of 4-n-alkoxybenzoyl chloride with 4-hydroxy benzaldehyde in dry pyridine and DMF.

The FT-IR spectrum of compound [I] showed many bands in the region (2981-2950) cm⁻¹ due to ν C-H aliphatic stretching, two sharp bands of carbonyl stretching (of ester and aldehyde groups) around 1728 cm⁻¹ and 1685 cm⁻¹, respectively, besides two stretching bands at 2846 cm⁻¹ and 2744 cm⁻¹ due to CH aldehydic. Finally, a sharp peak around 1269 cm⁻¹ is attributed of ν C-O stretching of ether (OR) group. المجلد 30 العدد (1) عام 2017

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Thiosemicarbazone [II] is synthesized by condensation of 4-(4'-methoxy benzoyloxy) benzaldehyde [I] and thiosemicarbazide in ethanol under reflux.

FTIR absorption-spectrum of this compound [II] exhibited the disappearance of aldehydic (C=O) and CH absorption bands together with appearance of new absorption stretching band at1695 cm⁻¹ which is assigned to vC=N stretching [17]. The spectrum showed many peaks in the region 3441-3088cm⁻¹ which could be attributed to asym. and sym. stretching vibration of NH and NH₂ groups [18] besides to two sharp peaks at 1732 cm⁻¹ and 1195cm⁻¹ due to ester and thion bonds, respectively.

Compound [III] was synthesized from the condensation of compound [II] with chloroacetic acid (in presence of sodium acetate). The FTIR spectrum showed the appearance of new stretching band due to a carbonyl group of thiazolidinone at 1690cm⁻¹[19]and stretching band appears at 698 cm-1 due to C-S group. While¹HNMR spectrum (in DMSO-d as a solvent), Figure (1) showed the following signals: singlet signal at δ (12.28) ppm [20] for one proton of CO- NH (amide) group. Many signals at δ (7.36-8.68) ppm that could be assigned to the eight aromatic protons, but signal at δ (8.1) ppm for a proton of imine, also a singlet signal at δ (3.59) ppm [21] for two protons at C-5 thiazolidinone ring. Finally a singlet signal appeared at δ (4.14) ppm for three protons of OCH₃ group.

The reaction of thiazolidenone [III] with acetic anhydride yielded a new compound [IV]. This compound is identified by FTIR spectroscopy. The FTIR spectrum of compound [IV] showed disappearance absorption band due to NH group and the appearance of a new band at 1699 cm⁻¹ due to stretching vibration carbonyl for amide group (COCH₃).

While, the compounds $[V]_{a,b,c}$ were synthesized by the reaction of one mole of compound [III] with two moles of amine with HNO₂ at 0-4 0 C to get diazonium salt, The FTIR spectrum of this compounds $[V]_{a,b,c}$ showed a stretching band at (1546-1504) cm⁻¹ [22] due to the N=N group, as in Figure (2) of compound $[V]_{c}$.

¹HNMR spectrum of compound $[V]_b$ (in DMSO) Figure (3) showed the following characteristics chemical shifts: a singlet signal at δ (3.99) ppm one protons at C-5 thiazolidinone ring, also a singlet signal at δ (4.12) ppm for three protons of OCH3 group and signals in the region δ (7.2-8.3) ppm that due to the eight aromatic protons, another singlet signal appeared at δ (8.35) ppm for a proton of imine, Also a singlet was observed at δ (12.28) ppm for one proton of CO-NH(amide) group.

The azo compound $[V]_c$ was first converted to ester compounds $[VI]_{a,b}$ by using benzoyl chloride or ansoylchloride in THF and triethyl amine. The ester was identified by FTIR spectra which showed the disappearance of a wide peak around (3442) cm⁻¹ which belongs to stretching vibration of the (-OH) group. The FT-IR spectra also, showed the appearance of the characteristic absorption band at (1732-1724) cm⁻¹[23] due to the stretching vibration of the (C=O) of the forming new ester group.

The 1,3-oxazepines[VII], [VIII], $[IX]_{a,b,c}$ were obtained by addition reaction of Schiff bases with naphthalic anhydride in dry benzene. The FTIR absorption bands of these compounds were confirmed from the disappearance of stretching band due to CH=N of Schiff bases with the appearance of two bands characteristic of two carbonyl groups of oxazepine ring in region (1760-1740) cm⁻¹ and (1697-1674) cm⁻¹[24], as in Figure (4) of compound [IX]_b.

The ¹HNMR spectrum of compound [VIII], Figure (5) showed twenty aromatic ring protons appear as multiplet in the range (δ 7.01-8.44) ppm, a singlet signal of N-CH proton of oxazepine absorbed at δ 6.87 ppm and two singlet signal at δ (2.3 and 2.2)ppm for two protons at C-5 thiazolidinone ring and COCH₃ group, respectively. Furthermore, a singlet signals at δ 4.14ppm for OCH3 protons absorbed.

Biological activity

These compounds stabilizer is reasonable biological activity due to presence of either NH or C=O and OH group sinits structure [25]. The cell wall structural nature of gram negative enteric bacteria may be responsible for the observed susceptibility. For instance,(the cell wall of gram negative bacteria contains 15-20% poly saccharides and 10-20% lipid, where gram positive bacteria contain 35-60% poly saccharides and only 0-2% lipid [26]. The poly saccharides and lipid contents of the cell wall affect the permeability of allicin and porrum constituents) [27, 28]. The effect of the prepared compounds was due to interfering with the structure of bacteria cell wall or by stopping bacteria multiplying [29]. All the oxazepine derivatives $[X]_{a-c}$ which were containing azo group show a moderate antibacterial activity against serratiamarcescens, while did not showed any antibacterial activity towards the other types Table (4).

Figures (6) exhibited the effect of the synthesized compounds on six types of bacteria.

References

- Jain, K. A.; Vaidya, A.; Ravichandran, V.; Kashaw, K. K. and Agrawal, k. R.(2012) "Recent development and biological activities of thiazolidinone derivatives", Bioorg. Med. Chem., xxx-xxx:1-18.
- 2. Gaikwad, S.; Suryawanshi,V.;Vijapure,Y. and Shinde, V.(2012) "Synthesis and biological study of some new thiazolidinone derivatives containing naphthofuran moiety", Journal of Chemical and Pharmaceutical Research, 4(3):1851-1855.
- Hamad, S. A. (2012) "Synthesis of some thiazolidinones and N- acetyl amino derivatives from 4-amino sulphamethaoxazole", DIYALA JOURNAL FOR PURE SCIENCES,8(4): 37-47.
- 4. Patel,D.; Kumari,P. and Patel,N.(2010) "Synthesis and characterization of some new thiazolidinones containing coumarin moiety and their antimicrobial study", Archives of Applied Science Research, 2 (6):68-75.
- Behbehani, H. and Ibrahim, H. (2012) "4-Thiazolidinones in Heterocyclic Synthesis: Synthesis of Novel Enaminones, Azolopyrimidines and 2-Arylimino-5-arylidene-4thiazolidinones", Molecules, 17: 6362-6385.
- 6. Kareem, A. and Ghanim, H. (2015) "Synthesis and Identification Some of 1,3-Oxazepine Derivatives", Journal of Applied, Physical and Biochemistry Research, 5(1): 45-56.
- 7. Sallal, Z. and Ghanem, H.(2011) "Synthesis of New 1,3-Oxazepine Derivatives Containing Azo Group", Journal of Kufa for Chemical Science, 2:11-23.
- Abood, H. Z. (2009) "Synthesis of Some New 1,3-Oxazepine and Tetrazole Derivatives Containing Azo Group", Journal of Kerbala University Scientific,7(1) :297-303.
- Sunil, D. ;Ranjitha, C. ; Rama, M. and KSR , P. (2014) "Oxazepine Derivative as an Antitumor Agent and Snail1 Inhibitor against Human Colorectal Adenocarcinoma", International Journal of Innovative Research in Science, Engineering and Technology, 3(8): 15357-15363.
- 10. AL-rammahi, A. ; AL-Khafagy, A. and AL-rammahi, F.(2015) "Synthesis and Characterization of Oxazepen and Imidazolin Derivatives From 2-Amino-5-

Mercapto-1,3,4- Thiadiazol and Studing of Their Biological Activity", World Journal of Pharmaceutical Research, 4(2):1668-1680.

- 11. Mustafa, A.(2015) "Novel synthesis and antibacterial activities of new derivatives of 7,8-Dichlorodibenzo(b, d)thiophene-2-carboxcyilic acid of pharmaceutical interest", Diyala Journal For Pure Sciences, 11(1):78-97.
- 12. Hamak, F. K. and Eissa, H. H.(2013) "Synthesis, Characterization, Biological Evaluation and Anti Corrosion Activity of some Heterocyclic Compounds OxazepineDerivatived from Schiff Bases", International Journal of ChemTech Research, 5(6): 2924-2940.
- 13. Pagariya, K.; Pathade, M. and Bodkhe, S.(2015)"Synthesis, Characterization and Antimicrobial screening of some Azo compounds derived from Ethyl vanillin", Research Journal of Chemical Sciences, 5(7): 20-28.
- 14. Aljamali, N. (2015) "Review in Azo Compounds and its Biological Activity", Biochem Anal Biochem.,4(2):1-4.
- 15. Abood, H. Z.; Haiwal, T. R.; ghanim, T. H. and Radhi, M. S. (2013) "Synthesis of Some New Schiff Bases, Tetrazole and 1,3-Oxazepine Derivatives Containing Azo Group and 1,3,4- Thiadiazole Moiety", Journal of Babylon University/Pure and Applied Sciences, 21(1):181-205.
- 16. Nogponge, K. W.; Aporn, M.; Duangti P. and Sukon, T.(2007) "Screening and identification of lactic acid bacteria producing antibacterial compound from pig gastrointestinal tract", Kmil Sci. Tech. J. ,8(1).
- 17. Mostafa, T. (2010) "Synthesis and Modification of some Heterocyclic Compounds with Potential Biological Activity Coupled on Poly (Maleic Anhydride –Methyl Methacrylate)", Journal of American Science, 6(8):512-524.
- 18. Muhsen, T.R.; Tomma, J. H. and Mukhlus, A. J. A. (2012) "Synthesis and Characterization of Novel Compounds Containing of Imidazolidinone and Oxazepine Rings", Ibn Al-Haitham J. Pur. App. Sci., 25(3): 276-283.
- 19. Sharma, R.; Samadhiya, P.; Srivastava, S. and Srivastava, S. (2012) "Synthesis and biological activity of 4-thiazolidinone derivatives of phenothiazine", J. Serb. Chem. Soc., 77 (1): 17–26.
- 20. Al-luhaibi, N. A.(2015), Ph.D. Thesis, College of Science, University of Mustansiriya.
- 21. Hassan, H. A. (2013) "Synthesis and Characterization of Some New 1,2,3-Triazole, Pyrazolin-5-one and thiazolidinone Derivatives", Journal of Al-Nahrain, University, 16 (1):53-59.
- 22. Karam, H. N.;Tomma, H. J. and AL-dujaili, H. A. (2013) "Synthesis and Characterization of Heterocyclic Compounds Derived From 4- Hydroxy and 4- Amino Acetophenone", Ibn Al-Haitham Jour. For Pure and Appl. Sci.,26(3) :285-301.
- 23. Jamel, M. N.; Hussein, F. D. and Tomma, H. J. (2014) "Synthesis and Characterization New Schiff Basea, Pyrazole and Pyrazoline Compounds Derived From Acid Hydrazide Containing Isoxazoline Ring", Ibn Al-Haitham Jour. For Pure and Appl. Sci., 27(3):435-447.
- 24. Saoud, A. S. (2011) "Synthesis and Characterization of Some New 1,3- Oxazepine Derivatives", Ibn Al-Haitham Jour. For Pure and Appl. Sci., 24(1) :1-9.

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- 25. Fahmy, M. M.; Mohamed, R. R. and Mohamed, N. A. (2012) "Novel antimicrobial organic thermal stabilizer and co-stabilizer for rigid pve", J. Molec ., 17 :7927-7940.
- 26. Carpenter, P. (1968) "Microbiology" 2nded..PhiladelphiaWB Saunders. 476
- 27. Nihal, T. Y.; Sedat, V.; Ferda, S. and Gokce, p. (2007) "Effect of extraction conditions on measured total polyphenol contents and antioxidant antibacterial activities of black tea", Molecules, 12:484-496.
- 28. Alzoreky, N. S. and Nakahara, K. (2003) "Antibacterial activity of extracts from some edible plants commonly consumed in Asia", Int. J. Food Microbial, 80:223-230.
- 29. Ahmad, I. and Beg. (2001) "Antimicrobial and phytochemical studies on 45 Indian medicinal plants against multiresistant human pathogens", J. Ethopharm., 74 :113-123.

Comp No.	Nomenclature	Structural formula	Molecular formula	M.P °C	Yield %	Color
[111]	4{[(4-oxothiazolidin-2-yli dene)hydrazono-methyl]phenyl} 4-methoxybenzoate		$C_{18}H_{15}N_3O_4S$	230-232	96	White
[IV]	4-{[(3-acetyl-4-oxothia- zolidin-2-ylidene) hydrazono -methyl]phenyl} 4- methoxybenzoate	CH ₃ O- CH ₃ O- CH ₃ O- CH ₃ O- CO ₂ CH ₃ O- CH ₃ O-	$C_{20}H_{17}N_3O_5S$	210-212	92	Off white
[V] _a	4-{[(4-oxo-5-(4-tolyldi- azenyl)thiazolidin-2-ylidene) hydrazono-methyl)phenyl 4- methoxybenzoate	CH_3O CO_2 CO_2 $CH:N-N \ll N=N-1$ CH_3 H O CH_3	C ₂₅ H ₂₁ N ₅ O ₄ S	195-197	91	Orange
[V] _b	4-{[(5-((4-methoxyphenyl) diazenyl)-4-oxothiazolidin-2- ylidene)hydrazono-methyl] phenyl} 4-methoxybenzoate	$CH_{3}O\sqrt{-CO_{2}-\sqrt{-}CH:N-N=\sqrt{S}_{M}} \sqrt{N=N-\sqrt{-}OCH_{3}}$	C ₂₅ H ₂₁ N ₅ O ₅ S	206-208	99	Brown
[V] _c	4-{[(5-((4-hydroxyphenyl)diazenyl)-4-oxothiazolidin-2- ylidene) hydrazono-methyl] phenyl}4-methoxybenzoate	$CH_3O \sim CO_2 \sim CH:N-N \ll N=N-\sqrt{O-OH}$	C ₂₄ H ₁₉ N ₅ O ₅ S	220-222	95	Black
[VI] _a	4-{[(5-((4-(benzoyloxy) phenyl)diazenyl)-4-oxothia- zolidin-2-ylidene)hydrazono- methyl] phenyl} 4-methoxy benzoate	$CH_3O\sqrt{-CO_2}$ CO_2 $CH=N-N=\sqrt{N-1}$ $NN-\sqrt{-O-C}$ $O-C$ V	C ₃₁ H ₂₃ N ₅ O ₆ S	170-172	96	Black
[VI] _b	 4-{[2-(4-(4-methoxybenzoyl- oxy)benzylidene)hydrazono)- 4-oxothiazolidin-5-yl)diazen- yl]phenyl} 4- methoxybenzoate 	$CH_{3}O\sqrt{-CO_{2}-CO_{2}-CH=N-N} \xrightarrow{S} NN \sqrt{-O-C} \xrightarrow{O} O-CH_{3}$	C ₃₂ H ₂₅ N ₅ O ₇ S	198-200	95	Brown

Table (1) The physical data of compounds [III]-[IX]

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[VII]	2-[4-(4-methoxybenzoyloxy) phenyl]-3-4-oxothiazolidin-2- ylideneamino)2,3- dihydrobenz [1,2e][1,3]oxazepin-4,7- diones	CH ₂ O CH	C ₂₆ H ₁₉ N ₃ O ₇ S	175-177	64	White
[VIII]	2-[4-(4-methoxybenzoyloxy) phenyl]-3(3-acetyl-4- oxothiazolidin-2- ylideneamino)2,3- dihydrobenz [1,2e][1,3]oxazepin-4,7- diones	$CH_{3}O$	C ₂₈ H ₂₁ N ₃ O ₈ S	169-170	63	Off white
[IX] _a	2-[4-(4-methoxybenzoyloxy) phenyl]-3-[4-oxo-5(4-tolyl) diazenyl)thiazolidin-2-ylidene amino)-2,3-dihydrobenz [1,2e] [1,3]oxazepin-4,7- diones	CH_3O CO_2 CO_2 $CH_{-N-N} $ $N=N-V$ CH_3 O H H O H O H	C ₃₃ H ₂₅ N ₅ O ₇ S	252-254	58	Pal brown
[IX] _b	2-[4-(4-methoxybenzoyloxy) phenyl]-3-[4-oxo-5(4- methoxyphenyl)- diazenyl) thiazolidin-2-ylidene amino]- 2,3-dihydrobenz [1,2e] [1,3]oxazepin-4,7-diones	$CH_{3}O \swarrow CO_{2} \swarrow O = CH-N-N = (N-N) = (N-N$	C ₃₃ H ₂₅ N ₅ O ₈ S	>290	42	Gray
[IX] _c	2-[4-(4-methoxybenzoyloxy) phenyl]-3-{5[(4- hydroxyphenyl) diazenyl)-4- oxo thiazolidin-2-ylidene amino]-2,3-dihydrobenz [1,2e] [1,3]oxazepin-4,7- diones	$CH_{3}O$ CO_{2} CO_{2} $CH-N-N \leq N-N = 0$ O H O H O H O	C ₃₂ H ₂₃ N ₅ O ₈ S	178-180	58	Dark brown

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Table (2) Characteristics FTIR absorption bands of compounds[V]_{a,b,c}-[VI]_{a,b}

Comp.N		Characteristics bands FTIR spectra(cm ⁻¹)							
	υNH	υ C-H aliphatic	υ C=O	υ C=C aromatic	υ N=N	Other			
$[V]_a$	3275	2980-2819	1722,1695	1600	1506	δCH ₃ 1371			
[V] _b	3257	2997-2837	1730,1700	1600	1505				
[V] _c	3296	2983-2810	1739,1718	1600	1546	υ OH: 3442			
[VI] _a	3290	2980-2850	1732-1680	1603	1504				
[VI] _b	3280	2970-2820	1724-1675	1598	1510				

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Comp.No.	Characteristics bands FTIR spectra(cm ⁻¹)						
	υNH	υC-H	υС-Н	υ C=O	υ C=C	υC-N	
		aromatic	aliphatic		aromatic		
[VII]	3323	3076	2966-2848	1755,1720,1693	1595	1356	
[VIII]	-	3070	2962-2833	1760,1728,1693	1604	1365	
[IX] _a	3320	3090	2960-2840	1756,1724,1695	1595	1355	
[IX] _b	3328	3080	2968-2820	1760,1722,1697	1597	1345	
[IX] _c	3310	3060	2895-2810	1740,1710,1674	1589	1340	

Key of symbols :

1.active(slightly) = + 5 - 10mm, 2. active(Moderately) = ++ 11 - 15mm3. active (Highly) = +++ more than 15mm.

Con: DMSO

Comp. No.	Serratiamarc escens	Staphylococcus- epidermidis	Bacillus	Pseudomonas- aeruginosa	klebseilla	providencea
[III]	++		++	-	++	-
[IV]	-		-			-
[V] _a	-	-	-	-	-	-
[V] _b	-		-	-	-	-
[V] _c	-	+++	-	-	++	-
[VI] _a	-	-	-	-	-	-
[VI] _b	-		-	-	H	-
[VII]	-	-	_	-	-	-
[VIII]	-	-	-	-	-	++
[IX] _a	++	-	-	-	-	-
[IX] _b	++		-	-	-	-
[IX] _c	++		-	+++	++	-
Con.	-	-	-	-	-	-

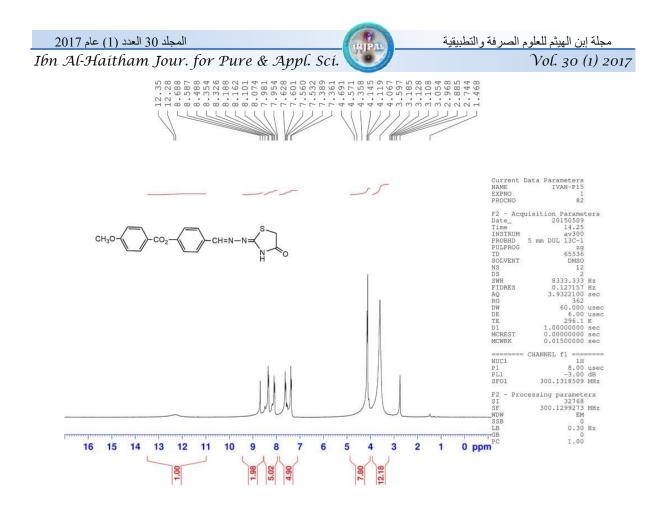


Figure (1) ¹HNMR-Spectrum of compound [III]

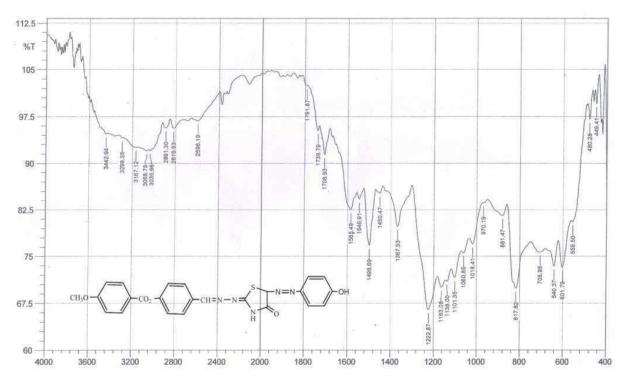


Figure (2) FTIR-Spectrum of compound [V]_c

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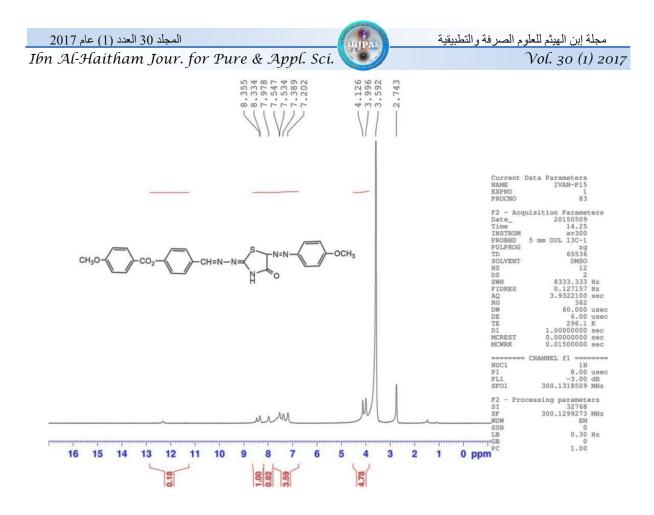


Figure (3) ¹HNMR-Spectrum of compound [V]_b

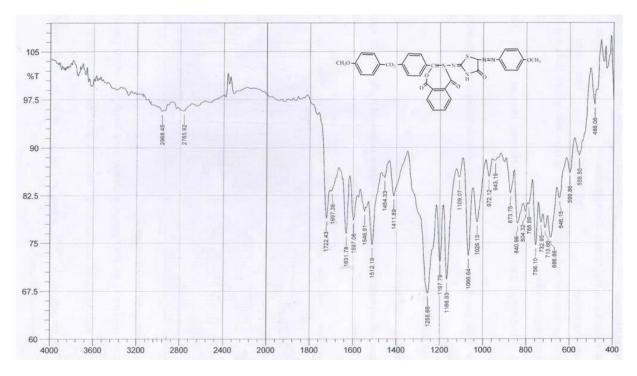


Figure (4) FTIR-Spectrum of compound [IX]_b

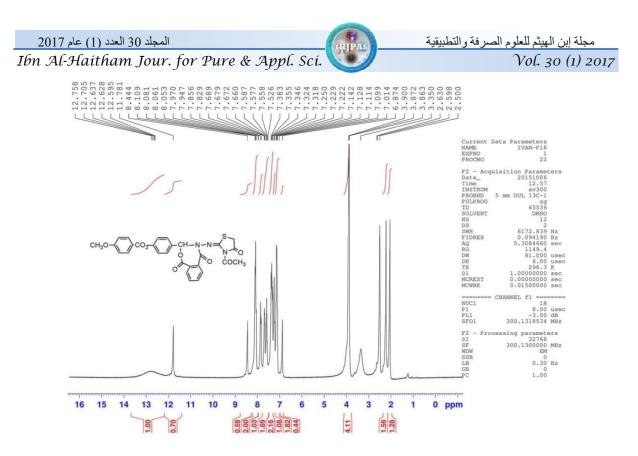


Figure (5) ¹HNMR-Spectrum of compound [VIII]



Figure (6) effect of compounds:[III]-[IX] on bacteria

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الخلاصة

حضر 4- ثايوزولدنون بثلاث خطوات من تفاعل انيسول كلورايد مع 4-هيدروكسي بنزالديهايد في خطوة يتكون خلالها الديهايد اروماتي[4- (`4-مثيوكسي بنزويل) بنزايل الديهايد[I] ومن ثم مفاعلة المركب الأخير مع ثايوسيمكاربزايد ليعطي ثايوسيمكاربازون [II] و بتفاعل ثايوسيمكياربازون مع كلورو حامض الخليك بوجود خلات الصوديوم يحضر مركب4- ثايزولدنون[III] الذي يستعمله مركب وسطي لتكوين مركبات جديدة حضر مركب [IV] من تفاعل [III] مركب. ثناعل تفاعل تايوسيمكياربازون مع كلورو حامض الخليك بوجود خلات الصوديوم يحضر مركب4- ثايزولدنون[III] الذي يستعمله مركب وسطي لتكوين مركبات جديدة حضر مركب [IV]من تفاعل [III] مع انهيدريد الخليك بينما مع الامينات ينتج مركب الازون مع كلورو حامض الخليك بوجود خلات الصوديوم يحضر مركب4- ثايزولدنون[III] الذي يستعمله مركب وسطي لتكوين مركبات جديدة حضر مركب [IV] مع انيسول كلورايد وبنزويل انهيدريد الخليك بينما مع الامينات ينتج مركب الازو مع الاوران وعند مفاعلة مركبات الازو مع انيسول كلورايد وبنزويل كلورايد في وسط قاعدي وسط قاعدي نصرت مركبات الازو مع انيسول كلورايد وبنزويل الهيدريد الخليك بينما مع الامينات ينتج مركب الازوه وساعي الاوراي من مركبات مركبات الازو مع انيسول كلورايد وبنزويل الهيدريد الخليك بينما مع الامينات ينتج مركب الازوم الاورالا والاورايد وند مماعلة مركبات الازو مع انيسول كلورايد وبنزويل [III] الهيدريد الخليك بينما مع الامينات ينتج مركب الازوم الورايد وبنزويل [III] وعند مفاعلة مركبات اوكسوزبين [IV] والالاورايد وينزويل الورايد في وسط قاعدي نصل على مركبات استرية جديدة إلاا]. حضرت مركبات اوكسوزبين [IV] والمركبات [IX] مع انهيدريد نفثالك في البنزين الجاف شخصت المركبات المحضرة باستعمال مطيافية إلى اللهالاليك البعض منها).

تم فحص المركبات المحضرة ضد ستة انواع من البكتريا وهي pseudomonas aeruginosa, klebseilla, تم فحص المركبات المحضرة ضد ستة انواع من البكتريا وهي aphylococcusepidermidis, و providencea, Serratiamarcescens (Gram negative) Bacillus(Gram positive).

الكلمات المفتاحية: ثايوزولدينون قواعد شف مركبات الازو13- اوكسوزبين والفعالية البكتيرية.