

YIELD AND QUALITY OF CHERRY TOMATO FRUITS IN HYDROPONIC CULTIVATION

PRODUÇÃO E QUALIDADE DE FRUTOS DE TOMATE CEREJA EM CULTIVO HIDROPONIA

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ABSTRACT: Because of the food and industrial importance of tomato, it holds great significance, and is one of the most produced species using the hydroponic cultivation systems. The objective of this study was to evaluate the effects of different concentrations of nutrient solution on the production and quality of cherry tomatoes (*Lycopersicon esculentum* ‘Samambaia’) grown in a hydroponic system in protected conditions. The experiment was conducted in pots filled with coconut fiber substrate using a randomized complete block design with four replications and six plants per plot. Five concentrations of nutrients were evaluated (50, 75, 100, 125, and 150% of the standard nutrient solution); the solutions produced the following electrical conductivities: 1.8, 2.0, 2.6, 3.4, and 3.9 dS m⁻¹, respectively. At 90 days after transplanting, the tomato fruits were harvested, at which time the production variables and post-harvest quality of mature fruits were determined. The best production and post-harvest quality indexes of cherry tomatoes (‘Samambaia’) were found when using 111% of the standard nutrient solution, corresponding to the concentrations of 9.44, 2.44, 2.22, 6.44, 4.11, 2.44, and 2.78 mmolc L⁻¹, of NO₃⁻, NH₄⁺, P, K, Ca, Mg, and S, respectively; and 66.6, 55.5, 14.4, 1.89, 0.56, and 0.44 mmol L⁻¹, of Fe, B, Mn, Zn, Cu, and Mo, respectively. Nutrient solutions with electrical conductivity above 2.89 dS m⁻¹ severely reduced the fruit yield of cherry tomatoes.

KEYWORDS: *Lycopersicon esculentum*. Nutrients. Irrigation. Hydroponics system.

INTRODUCTION

The growing demand for increased productivity and quality of agricultural products is a catalyst for agricultural development, thus, there is a constant need to adapt products and the means of production. This aspect is particularly relevant to horticulture because most producers have a small growing area in which they seek to obtain the maximum possible income (RINALDI et al., 2008). Soilless crops allow production in a small area and can maintain product quality at satisfactory levels, as well as serving as a strategy for soil conservation and preservation of water sources (OLIVEIRA et al., 2014, OLIVEIRA et al., 2016).

Several authors have verified that plants respond differently to the same salinity conditions of nutrient solution when cultivated in soil and hydroponics. In soil cultivation, the moisture level varies between irrigations, decreasing the osmotic and matric potentials. However, in hydroponic cultivation, the potassium potential is practically null due to the saturated state in which the plant is

growing. This is an advantage when only saline waters are available to prepare the nutrient solutions. Therefore, with similar salinity conditions, hydroponic cultivation allows a greater absorption of water and nutrients by the plants, with a lower energy expenditure in comparison to soil cultivation (SILVA et al., 2013a; SÁ et al., 2015; SANTOS et al., 2017).

Furthermore, hydroponic crops make it possible to obtain better quality products when compared to conventional systems. Indoor hydroponic crops offer greater uniformity of crop/fruit, better water use efficiency, as well as the ability to control other factors during production (GENÚNCIO et al., 2006; CARDOSO et al., 2017; SANTOS et al., 2017).

The formulations of the nutrient solutions used in the hydroponic production of plants are of great interest. Several formulations of nutrient solution are proposed in the specialized literature, however, it is necessary to determine the effects of solution concentration and nutrient ratios on plant growth, plant development, production, and fruit

quality (FERNANDES; MARTINEZ; OLIVEIRA, 2002; MELO et al., 2014; DINIZ et al., 2015).

In Brazil there has been significant growth in the tomato growing sector, with a particular focus on hybrid varieties, especially those within the cherry tomato group (GUILHERME et al., 2014). This demand has partly been driven by the market, as the use of cherry tomatoes as adornments and appetizers in the cooking has become more popular. Besides the values of market, has been more interested farmers because they spread their cultivation in a protected environment (CARVALHO; TESSARIOLI NETO, 2005).

Vegetables respond very well to hydroponic cultivation and most of the water sources of the Brazilian semi-arid region are saline. The objective of this study was to evaluate the effects of the application of different concentrations of nutrient solution on the production and quality of cherry tomato fruits cultivated in a hydroponic system in protected environment conditions.

MATERIAL AND METHODS

The experiment was carried out in a chapel-type greenhouse with a ceiling height of 3.0 m, 20 m length, and 14 m width, located in the Department of Environmental Sciences of the Federal Rural Semi-Arid University (UFERSA), Mossoró, State of Rio Grande do Norte, Brazil (5°11'31" S, 37°20'40" W). The average altitude of the site is 18 m. According to Köppen's climatic classification, the climate of the region is BSw_h, with an average annual temperature of 27.3°C, fairly irregular rainfall with an annual average of 672.9 mm, and a relative humidity of 68.9%.

A randomized block design, with four replications, and six plants per plot, was used as the experimental design. The effects of five nutrient solution concentrations were evaluated: T₁ = 50, T₂ = 75, T₃ = 100, T₄ = 125, and T₅ = 150% of the standard nutrient solution recommended by Furlani et al. (1999) for tomato culture under tropical conditions.

The composition of the macronutrients in the solutions were: T₁ = 4.25, 1.1, 1.0, 2.9, 1.85, 1.1 and 1.25; T₂ = 6.375, 1.65, 1.5, 4.35, 2.775, 1.65 and 1.875; T₃ = 8.5, 2.2, 2.0, 5.8, 3.7, 2.2 and 2.5; T₄ = 10.625, 2.75, 2.5, 7.25, 4.625, 2.75 and 3.125; T₅ = 12.75, 3.3, 3.0; 8.7, 5.55, 3.3 and 3.75 mmol L⁻¹, of NO₃⁻, NH₄⁺, P, K, Ca, Mg and S, respectively, according to the recommendation for tomato, cited by Furlani et al. (1999). To prepare the nutrient solution, calcium nitrate, potassium nitrate, monoammonium phosphate (MAP), and magnesium

sulphate were added to 250 L of tap water (EC = 0,6 dS m⁻¹) in the following amounts respectively: T₁ (95.62, 88.35, 36.9, 53.75 g); T₂ (143.44, 132.52, 55.35, 80.62 g); T₃ (191.25, 176.70, 73.80, 107.5 g); T₄ (239.06, 220.87, 92.25, 134.37 g); and T₅ (286.87, 265.05, 110.7, 161.25 g).

The composition of the micronutrients in all solutions was 60, 50, 13, 1.7, 0.5, and 0.4 mmol L⁻¹, of Fe, B, Mn, Zn, Cu and Mo, respectively, according to the recommendation for tomato, cited by Furlani et al. (1999). Also, the following micronutrients were added (M): 4.41 g of B-H₃BO₃, 0.58 g of Cu-CuSO₄, 3.13 g of Mn-MnSO₄, and 0.37 g of Mo-Na₂.MoO₄.2H₂O and, as source of iron (I) 33.85 g of Fe7 was added. The combinations of M and I used in the nutritive solutions were as follows T₁: 50 mL of M and 50 mL of I; T₂: 75 mL of M and 75 mL of I; T₃: 100 mL of M and 100 mL of I; T₄: 125 mL of M and 125 mL of I and T₅: 150 mL of M and 150 mL of I.

The physicochemical composition of the tap water used in the preparation of the nutrient solution was as follows: EC = 0.60 dS m⁻¹; pH = 8.0; Ca²⁺ = 0.6 (mmol L⁻¹); Mg²⁺ = 0.1 (mmol L⁻¹); Na²⁺ = 5.1 (mmol L⁻¹); Cl⁻ = 1.8 (mmol L⁻¹); CO₃²⁻ = 0.5 (mmol L⁻¹); HCO₃⁻ = 3.8 (mmol L⁻¹); and RAS = 8.62 (mmol L⁻¹)^{0.5}. After the addition of the nutrients to the tap water, nutritive solutions were produced with electrical conductivities (EC) of T₁ = 1.8, T₂ = 2.0, T₃ = 2.6, T₄ = 3.4, and T₅ = 3.9 dS m⁻¹, respectively. To correct the pH of the solutions to values around 6.5, sulfuric acid was added to the nutrient solution.

Each experimental plot was composed of an individual hydroponic system, consisting of 6 plastic pots of 11 L, with spaces of 0.5 m between pots and 1.0 m between lines. The pots had holes drilled in their bases to drain excess water. The pots were filled with coconut fiber and placed on a support 0.10 m from the soil level of the greenhouse to avoid direct contact of the pot with the floor and to facilitate drainage. The plants were vertically tutored with the aid of string, and the other cultural treatments were carried out following the recommendations of Filgueira (2008).

A localized irrigation system with micro-tube emitters of 1.5 mm internal diameter was adopted. The nutrient solution was supplied through individual reservoirs with a capacity of 300 L each, suspended on a fixed iron structure at a height of 1.2 m. The volume applied in each irrigation was determined according to plant water needs based on the water balance in the root zone. This was obtained from the difference between the applied volume and the volume drained in the previous

irrigation, in extra pots arranged in the experimental area, to avoid losses of nutrients by leaching.

The cherry tomato seedlings ('Samambaia') were produced in trays of 180 cells with 1 tomato seed per cell, and irrigated twice a day with tap water ($EC = 0.6 \text{ dS m}^{-1}$). The seedlings were transplanted to the experimental units after 23 days. The tomato plants 'Samambaia' of determined growth habit. The plants were conducted with only one stem. The reproductive phase was started at 45 days, all the mature fruits were harvested between 70 and 90 days after transplanting to evaluate the yield, being determined the variables total production per plant, number and mean mass of the fruits. Samples of fruits with similar characteristics were taken to the Post-Harvest Laboratory of the Federal Rural Semi-Arid University (UFERSA) for analysis of total titratable acidity, total soluble solids, hydrogenation potential, and vitamin C content.

The titratable acidity was determined by titration using an aliquot of 1.0 mL of extracted, to which 49.0 mL of distilled water and 3 drops of 1% alcoholic phenolphthalein were added, using sodium hydroxide solution (NaOH) 0.1 N, standardized with potassium biphthalate, as titrant (IAL, 2008). The soluble solids were determined directly in the homogenized juice using a digital refractometer (model PR - 100, Palette, AtagoCo LTD., Japan), according to the recommendation proposed by the Association of Official Analytical Chemists (AOAC, 2002). The hydrogenation potential (pH) was determined using a pH meter (Model mPA - 210P / Version 7.1, Tecnoyon, Brazil), with direct insertion of the electrode, according to IAL (2008). Following the methods of AOAC (2002), the vitamin C content was determined by titration with 2,6-dichlorophenolindophenol (DFI) until a permanent clear rose coloration was obtained, using 1.0 mL of the diluted juice in 49.0 mL of oxalic acid 0.5%.

Data were submitted to analysis of variance, the means of the results were compared by the F test (5%) and submitted to polynomial regression analysis, with the software SISVAR 5.3 (FERREIRA, 2014).

RESULTS AND DISCUSSION

The proportion of nutrients in the nutrient solution (NS) influenced the production of fruits per cherry tomato plant in a quadratic form, with an estimated maximum production of 258.48 g in the nutrient solution concentration of 101.88%; above this concentration there is a decrease in production

(Figure 1A). In general, the yield values of tomato varied from 127.62 to 283.62 g in function of the concentrations in the nutrient solution (50, 75, 100, 125 and 150%). It can be inferred that fruit yield was negatively affected by the salinity of the hydroponic nutrient solution above 2.6 dS m^{-1} , this was probably because of the deleterious effects of salinity caused by excess nutrients in solution.

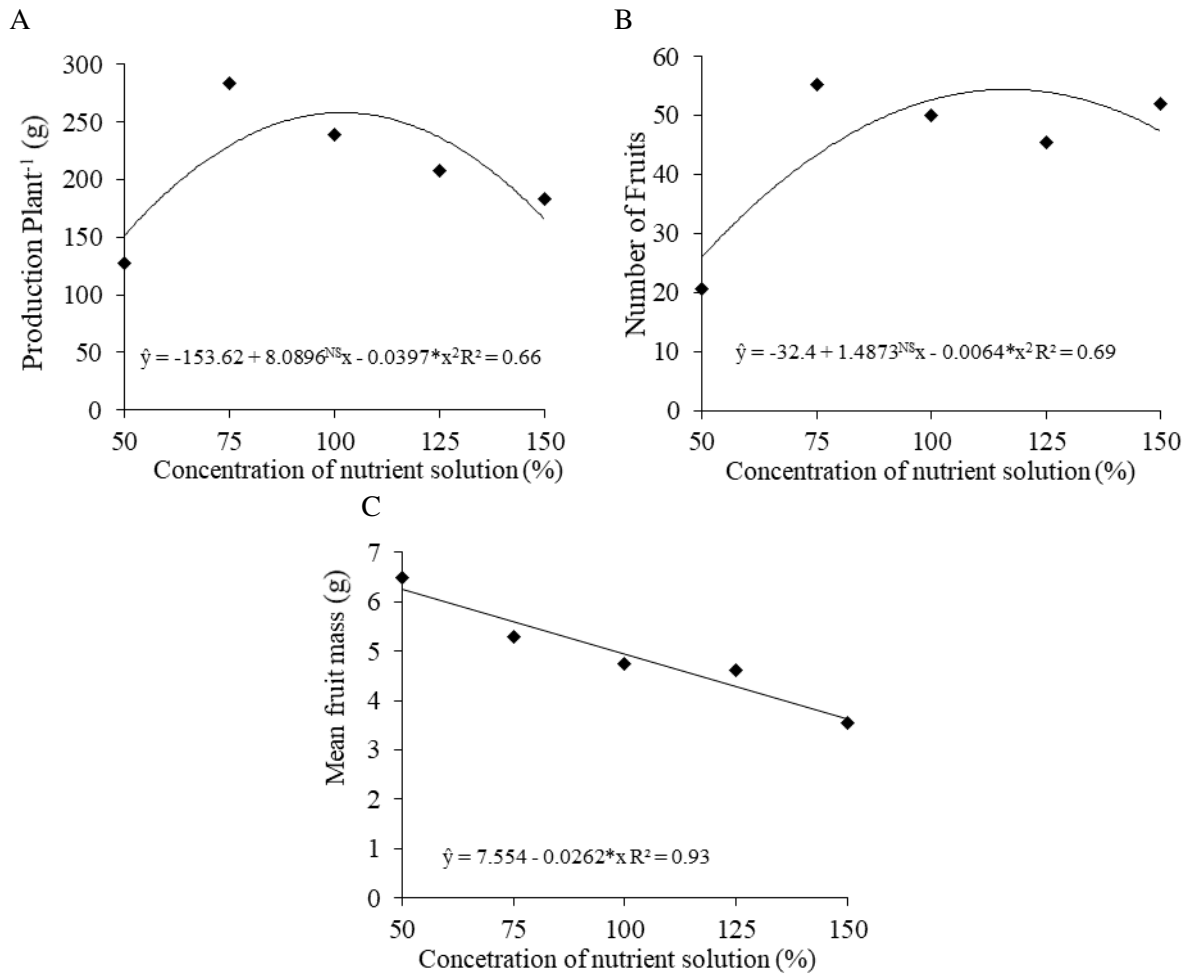
The mean yield of fruits per plant found were higher than those reported by Silva et al. (2012) in experiments conducted in the open field and when irrigated with different salinity levels of irrigation water (0.8, 1.6, 2.4, 3.2, and 4.0 dS m^{-1}), which recorded average fruit weight of 70.30, 101.54, 101.63, 123.50, and 155.79 g, respectively. On the other hand, Gomes et al. (2011) observed a linear reduction in the production of cherry tomato fruits in a hydroponic system using desalination. Furthermore, Medeiros et al. (2011) reported that salinity reduced the availability of water and nutrients to the plants, causing losses of average fruit weight as the saline concentration increased. Cosme et al. (2011) working with nutritional solution prepared with the addition of reject saline observed a reduction in the production of tomato fruits. The authors attributed this reduction to the increased salinity of the nutrient solution.

The increase of nutrient ratios in the nutrient solution significantly influenced the number of tomato fruits plant⁻¹ ($p < 0.01$). The number of fruits ranged from 45.5 to 55.25 fruits per plant, and increased with increasing nutrient solution concentration (Figure 1B), with a maximum number of fruits of 54.01 in 116.19% concentration of standard nutritional solution. These values were higher than those obtained by Silva et al. (2012) in cherry tomato which were from 4.87 to 8.75 fruits per plant at saline concentrations of 0.8 to 4.0 dS m^{-1} . These results may be associated with the plant response to nutrient solution concentrations, in other words, the high saline concentration of the nutrient solution causes a reduction in water absorption due to the osmotic effect and may also interfere with the availability of other nutrients and, consequently, causes reduction in the growth and the development of the plants (ROCHA et al., 2010; STRASSBURGER et al., 2011; CARDOSO et al., 2017; SANTOS et al., 2017).

The mean mass of cherry tomato fruit was linearly reduced by increasing the nutrient concentration in the nutrient solution by 0.66 g for a 25% increase in nutrient concentration (Figure 1C). Initially the reduction of the average mass of fruits was due to the increase in the number of fruits per plant. The plants grown with nutrient solution

concentrations of 75 and 100% had double the number of fruits compared to the plants grown with 50% (Figure 1B). Plantas cultivadas em 125 e 150% de solução nutritiva tiveram menos frutos do que as plantas cultivadas com 100% de solução, e seus frutos pesaram menos em média. Isso pode estar diretamente relacionado aos efeitos de salinidade da solução nutritiva (Figura 1C). Cosme et al. (2011) studied the effects of salinity of nutrient solution on

tomato plants, and verified that the growth and fruit production variables reduced significantly at high salinity levels. Other studies on the effects of salinity on plants have shown morphological alterations of the plants are common, resulting in reductions in growth, and changes to physiology and plant production (ROCHA et al., 2010; STRASSBURGER et al., 2011; SILVA et al., 2013a; DINIZ et al., 2015; SÁ et al., 2015).



* and ^{NS} = significant at 5% probability and not significant, respectively.

Figure 1. Production (A), number of fruits (B) and mean mass (C) of cherry tomato fruits cv. Samambaia in hydroponic cultivation as a function of nutrient ratios in the nutrient solution.

The titratable acidity of tomato fruits was altered by nutrient ratios in the nutrient solution (50, 75, 100, 125 and 150%), with values varying from 16.25 to 18.9%. The highest average value estimated was 18.18% fruit⁻¹, corresponding to 116.6% of the standard nutrient solution, with subsequent decreases above this nutrient concentration level (Figure 2A). The results are lower than the variation obtained by Feltrin et al.

(2005) when evaluating the productivity and fruit quality of fertirrigated tomato cultivars with chloride and potassium sulphate in soil covered with black plastic mulch in a greenhouse, which verified that the means of titratable acidity did not suffer to the effects of mineral nutrition. However, there was a significant difference in titratable acidity between the cultivars, where the values ranged from 0.57% to 0.80%.

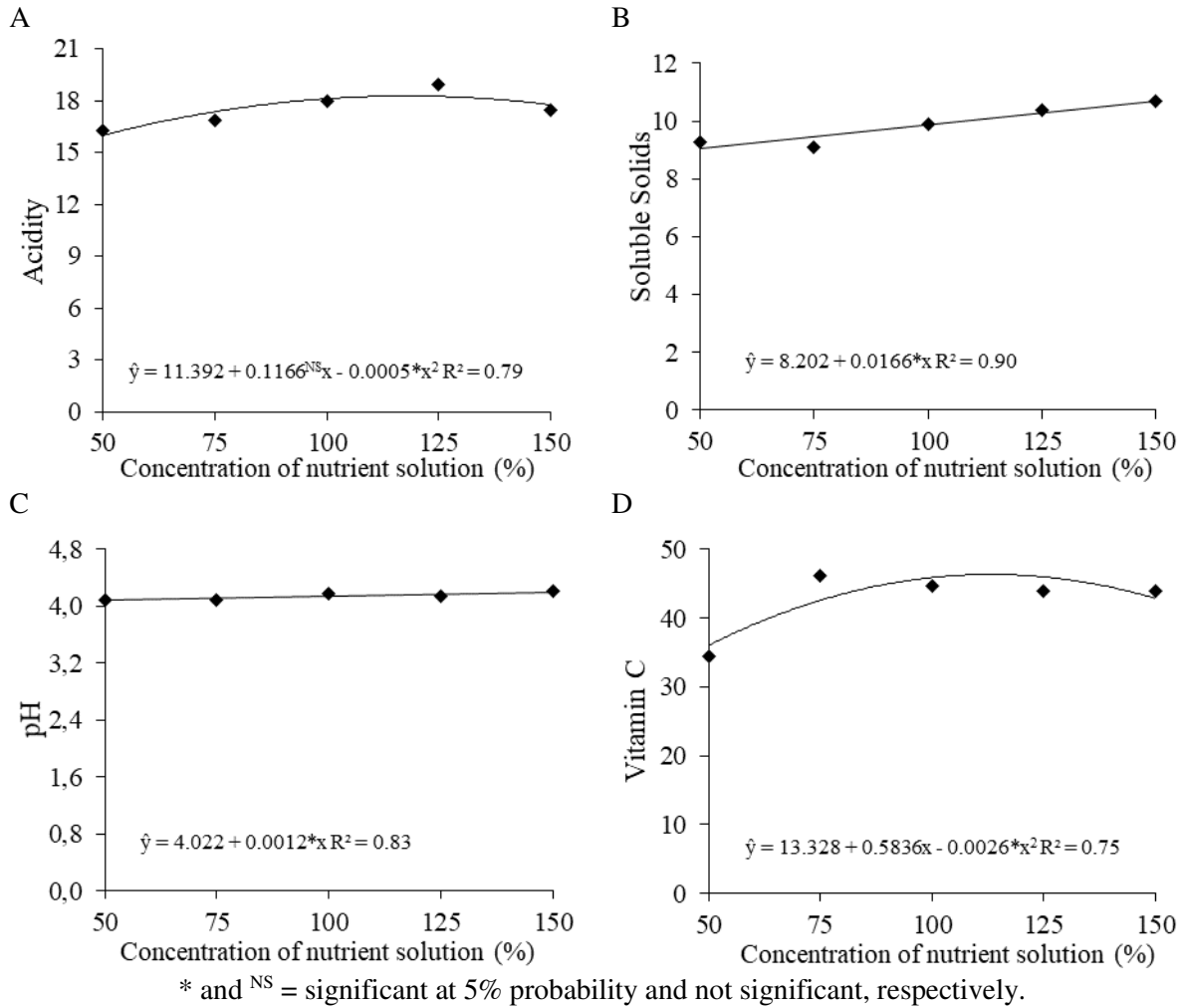


Figure 2. Acid titratable, % citric acid (A), soluble solids, Brix (B), hydrogenation potential values, pH (C) and vitamin C, mg of ascorbic acid/100 g (D) of cherry tomato fruits cv. Samambaia in hydroponic cultivation as a function of nutrient ratios in the nutrient solution.

Soluble solids contents (Brix) increased linearly as a function of the nutrient concentration increase in the nutrient solution, registering 0.42 Brix for each 25% increase in solution concentration (Figure 2B). This can be explained by the excess salts in the nutrient solution due to the increase of nutrient concentrations (1.8, 2.0, 2.6, 3.4 and 3.9 dS m⁻¹), which increased the ionic concentration in the interior of the plants and, consequently, the export of these ions to the fruits (ROCHA et al., 2010). The results obtained in the research were superior to those presented by Silva et al. (2012) on cherry tomatoes irrigated with high-salinity water (4 dS m⁻¹), which recorded linear increases in soluble solids content (°Brix) up to 7.98.

The application of different nutrient ratios in the nutrient solution had a significant effect (p < 0.01) on the pH values of cherry tomato fruits (Figure 2D). The pH of the cherry tomato fruit pulp increased from 4.09 to 4.21 at nutrient ratios of 50

to 150%, respectively (Figure 2C). Fruit pH also increased as the salinity of the nutrient solution increased (1.8, 2.0, 2.6, 3.4 and 3.9 dS m⁻¹). The results obtained for pH were similar to those observed by Feltrin et al. (2005) who evaluated the quality of fruits of fertirrigated tomato cultivars with chloride and potassium sulphate, and showed pH values from 3.96 to 4.17.

As observed for the other characteristics of fruit quality, the increase of the nutrient ratios in the nutrient solution significantly influenced the vitamin C content of the cherry tomato fruits; it had a verified quadratic behavior with maximum vitamin C in the fruit of 46.07 (mg of ascorbic acid/100 g of fruit) obtained in cultivated plants at the estimated concentration of 112.23% (Figure 3E). The levels of vitamin C observed in the present study exceed the range of 13.4 - 36.9 (mg of ascorbic acid/100 g of fruit) observed by Munhoz et al. (2011) and Silva et al. (2013b). This indicates that the cherry tomato has

higher concentrations of vitamin C than the conventional tomato, which may be related to its reduced size.

CONCLUSIONS

The best production and post-harvest quality indexes of cherry tomatoes (cv. Samambaia) cultivated in a hydroponic system were found when using 111% of the standard nutrient solution,

corresponding to the concentrations of 9.44, 2.44, 2.22, 6.44, 4.11, 2.44 and 2.78 mmol_e L⁻¹, of NO₃⁻, NH₄⁺, P, K, Ca, Mg and S, respectively, and 66.6, 55.5, 14.4, 1.89, 0.56 and 0.44 mmol_e L⁻¹, of Fe, B, Mn, Zn, Cu and Mo, respectively.

Nutrient solutions with electrical conductivity above 2.89 dS m⁻¹ promote reductions in the production of cherry tomatoes grown on coconut fiber substrates under protected conditions.

RESUMO: Devido à importância alimentar e industrial do tomateiro, a cultura destaca-se entre as hortaliças, sendo uma das espécies mais produzidas em sistema hidropônico de cultivo. Nosso objetivo foi avaliar os efeitos da aplicação de diferentes concentrações de solução nutritiva na produção e na qualidade dos frutos de tomate cereja (*Lycopersicon esculentum*, cv. Samambaia) em sistema hidropônico sobre condição de ambiente protegido. O experimento foi conduzido em vasos preenchidos com substrato de fibra de coco utilizando o delineamento de blocos casualizados, com quatro repetições e seis plantas por parcela. Foram avaliadas cinco concentrações de nutrientes na solução nutritiva hidropônica (50, 75, 100, 125 e 150% da solução nutritiva padrão) que, após a diluição dos nutrientes em água de torneira, estas produziram as seguintes condutividades elétricas: 1.8, 2.0, 2.6, 3.4 e 3.9 dS m⁻¹, respectivamente. Aos 90 dias após o transplântio, os frutos de tomate foram colhidos, ocasião em que se determinaram as variáveis de produção e qualidade pós-colheita de frutos maduros. A análise dos resultados indicou que os melhores índices de produção e qualidade dos frutos de tomate cereja cultivado em sistema hidropônico foram encontrados nas concentrações de macronutrientes da solução nutritiva correspondente a 9,44; 2,44; 2,22; 6,44; 4,11; 2,44 e 2,78 mmol_e L⁻¹ de NO₃⁻, NH₄⁺, P, K, Ca, Mg e S, respectivamente, e 66,6; 55,5; 14,4; 1,89; 0,56 e 0,44 mmol_e L⁻¹, de Fe, B, Mn, Zn, Cu e Mo, respectivamente. As soluções nutritivas com condutividade elétrica acima de 2,89 dS m⁻¹ reduziram severamente o rendimento de frutos de tomate cereja.

PALAVRAS-CHAVE: *Lycopersicon esculentum*. Nutrientes. Irrigação. Hidroponia.

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