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# Molecular docking studies of salubrinal and its analogs as inhibitors of the GADD34:PP1 enzyme

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## Abstract

The phenomenon of the endoplasmic reticulum (ER) stress as a molecular pathophysiological process underlies diseases as cancer, diabetes mellitus, myocardial infarction, neurodegenerative disorders, diseases of the urinary system, disorders associated with bone integrity, etc. To prevent ER stress, salubrinal, which is a phosphatase inhibitor of the eukaryotic translation initiation factor -GADD34:PP1, is currently being intensively studied. The aim of this work is to search for new analogues of this drug using molecular docking methods. Optimization of the geometry of the studied structures and molecular docking was carried out using the ArgusLab 4.0.1 software package. The three-dimensional crystal structure of the GADD34: PP1 enzyme (PDB ID: 4XPN) was loaded in the PDB format from the protein molecule data bank. The model of the binding site was created on the basis of the phosphoric acid residue (403 PO4). The dimensions of the binding site were set manually and were 40.000 Å along the X-axis, 40.000 Å - the Y-axis and 40.000 Å - the Z-axis. The docking was done with a flexible ligand, and the semi-empirical AScore function was used for the scoring procedure. It was shown that for the salubrinal molecule the most favorable was the conformation stabilized by the intramolecular hydrogen bond formed between the hydrogen atom of the thiourea fragment and the oxygen atom of the amide fragment. According to molecular docking data, six compounds from the fifty-four analyzed analogues of salubrinal exceed it in the stability of the complex formed with GADD34:PP1. The results of this work can be used to create new phosphatase inhibitors of the eukaryotic translation initiation factor GADD34:PP1.

#### Keywords

salubrinal; molecular docking; GADD34:PP1; RMSD; endoplasmic reticulum stress

#### Introduction

Endoplasmic reticulum (ER) is an intracellular membrane organelle that is extremely sensitive to changes in homeostasis. The membrane ER is integrated with the cell nucleus membrane. The internal ER space opens directly into the perinuclear space, which accompanies the contact of the ER signalling device with the genetic material. There are granular (rough) ER and agranular (smooth) ER. Smooth ER is located on the periphery of the organelle and is responsible for the synthesis of lipids, steroids, the metabolism of carbohydrates, medicines and other exogenous products [1,2].

Rough ER is an extension of the cell nucleus membrane. On its cytosolic surface, ribosomes are deposited, which provide for the translation of the protein directly into the ER cavity through the system of transmembrane channels. Inside the granular ER, "immature" protein molecules are foldable, i.e. take a correct spatial conformation. All unfolded or incorrectly folded proteins are caught and necessarily destroyed. Accumulation of misfolded protein molecules results in a functional overload of ER. This phenomenon is called ER stress and it leads to disorders in the normal functioning of the cell and threatens it with death [1,2].



Figure 1. Schematic simplified image of the main signal-sensory systems of ER stress

Over the past 15 years [3], molecular mechanisms of ER stress have been intensively studied as a fundamental phenomenon of cell protection from the action of various factors and as a molecular pathophysiological process leading to many severe diseases [3-21].

Figure 1 schematically depicts the response of the cell to ER stress, which is necessary for the cell to find ways to escape from the state of stress caused by the accumulation of unfolded or misfolded proteins, and which is mediated by three signal-sensory systems that begin in the ER lumen and terminate in the cytoplasm and nucleus [22-26]. Induction of ER stress stops the penetration of synthesized proteins into it and accompanies both the proper folding of proteins, which are already in it and the degradation of misfolded ones. This is necessary for the survival of the cell under the conditions of the factors that induce this stress, or death of the cell through the apoptosis system associated with ER [27,28].

The main signal-sensory ER stress systems (PERK, ATF6 and IRE1), which originate in its lumen under the conditions of accumulation of unfolded or incorrectly folded proteins in it, initiate total repression of translation initiation by phosphorylation of eukaryotic translation initiation factor  $2\alpha$ (eIF2 $\alpha$ ), and activation of the transcription of stress dependent genes by the formation of an active form of transcription factors ATF4 and ATF6, as well as an alternative splice variant of the transcription factor XBP1 (X-Box Protein-1), which controls the expression of the cell genes [19,26]. In this way, the response of cells to ER stress, which is mediated by the three signal-sensory systems, is necessary for the cell to find possible ways out of the state of stress caused by the accumulation of unfolded or incorrectly folded proteins in the ER lumen.

 $EIF2\alpha$  is a key participant in protein translation because it is responsible for binding the 40S ribosomal subunit to  $tRNA_{imet}$  (initiation of methionine tRNA), which recognizes the mRNA start codon and starts the synthesis of the peptide chain [19]. PERK phosphorylates  $eIF2\alpha$  translating it into an inactive  $eIF2\alpha P$  form. However, the holoenzyme complex GADD34: PP1 dephosphorylates  $eIF2\alpha P$ , again translating it into an active  $eIF2\alpha$  form.

In 2005, M. Boyce and colleagues reported that salubrinal (Fig. 2) acted as a phosphatase inhibitor GADD34:PP1, selective for eukaryotic translation initiation factor  $2\alpha$  (eIF2 $\alpha$ ) [29]. Thus, salubrinal weakens the synthesis of unfolded or misfolded proteins contributing to the preservation of homeostasis in ER and saving cells from apoptosis.



Figure 2. Structure of the salubrinal molecule

Since the beginning of intensive studies of salubrinal, its protective effect has been confirmed in a number of studies [30]. Although salubrinal is currently under development, we can already say with certainty about its prospects in the treatment of diabetes [31], myocardial infarction [32], neurodegenerative disorders [33,34], oncological diseases [35], diseases of the genitourinary system [36] and disorders related to the integrity of bone tissue [37,38]. Work is underway to study its toxicity and the development of analogues [39].

In this paper, using the methods of molecular docking [40,41], we have established the binding site of the salubrinal preparation with holoenzyme GADD34:PP1 and searched for analogues of this drug.

## Materials and methods

## Computer specification

All calculations were carried out on a Toshiba personal computer, the Satellite L650D model, AMD Phenom(tm) II P820 Triple-Core Processor. A 64-bit operating system was used.

## Ligand preparation

The search for structures for research was conducted in the SciFinder database (https://scifinder.cas.org) (see supporting information). Prior to molecular docking, the structures of all the compounds studied were optimized within the semiempirical PM3 method [42] using the ArgusLab 4.0.1 software package [43-47]. The calculation of the electron density distribution in the static salubrinal molecule was carried out with the ZINDO approximation method [48] in the same software package.

#### Protein preparation

The three-dimensional crystal structure of the GADD34:PP1 enzyme (4XPN) was loaded in the PDB format from the protein molecules data bank (<u>http://www.rcsb.org</u>). Prior to docking, the molecules

of all the non-proteinaceous components, except for one phosphoric acid residue, having the code in co-crystallisate 403 PO4, were removed. Hydrogen atoms were added throughout the protein structure before molecular docking.

#### Molecular docking procedure

Based on the phosphoric acid residue (403 PO4), a ligand group was created with the given name Ligand\_X-ray. Based on this group, a three-dimensional model of a binding site was created, the dimensions of which were set manually and amounted along the X-axis – 40.000 Å, the Y-axis – 40.000 Å and the Z-axis – 40.000 Å. Docking was performed with a flexible ligand. For the scoring procedure, the semi-empirical function AScore was used created on the basis of the XScore function [49]. The resolution of the cell was set at 0.250 Å. The calculation type was Dock; Docking Engine - ArgusLab. Visualization of the results was carried out using the program PyMOL [50].

### **Results and discussion**

## The results of ligand geometry optimization

According to the results of optimization of the geometry of the salubrinal molecule, the most stable is the conformation stabilized by the intramolecular hydrogen bond formed between the hydrogen atom of the thiourea fragment and the oxygen atom of the amide fragment (Fig. 3a). The length of the NH...O=C bond is 1.891 Å. That is, the salubrinal molecule exists as a pseudo 1,3,5-oxadiazine ring with an angle H...O=C 108.33 °. According to X-ray diffraction data for 1,3,5-oxadiazine cycles, this angle is somewhat larger and lies within the range of 114.76-120.00° [51-53]. The appearance of an intramolecular hydrogen bond is obviously associated with a large difference in the static charges on the oxygen atom of the amide fragment and the hydrogen atom of the thiourea fragment. According to calculations of the electron density distribution in the static salubrinal molecule (the ZINDO approximation method), on the oxygen atom  $\delta^-$  lies within -0.0409 $\rightarrow$ -0.0500, in turn, on the hydrogen atom  $\delta^+$  is 0.0500 $\rightarrow$ 0.0409 (Fig. 3b). The presence of an intramolecular hydrogen bond is characteristic of all the salubrinal analogues studied (*see* Supporting information, Tables S1 and S2).



Figure 3. a) The calculated structure of the salubrinal molecule (PM3 method), visualization in PyMol; b) Distribution of electron density in the static salubrinal molecule. Colors: a) □ 0.0500  $\rightarrow$  0.0409; b) = 0.0409  $\rightarrow$  0.0318; c) = 0.0318  $\rightarrow$  0.0227; d) = 0.0227  $\rightarrow$  0.0136; e) = 0.0136  $\rightarrow$  0.0045; f) = 0.0045  $\rightarrow$  - 0.0045; g) = -0.0045  $\rightarrow$  -0.0136; h) = -0.0136  $\rightarrow$  -0.0227; i) = -0.0227  $\rightarrow$  -0.0318; j) = -0.0318  $\rightarrow$  -0.0409; k) = -0.0409  $\rightarrow$  -0.0500

## The results of molecular docking

The active center of selective dephosphatase of the eukaryotic translation initiation factor (Fig. 4a) contains the phosphoric acid residue, and two Mg<sup>2+</sup> ions (not shown in the figure). In the active site of the GADD34:PP1, it is possible to distinguish three sites, one hydrophilic - located approximately in its center, and two lipophilic ones located on the periphery. Therefore, the interactions of the salubrinal

molecule with the active site of the GADD34:PP1 enzyme are represented by both polar contacts (Fig. 4b) and lipophilic interactions between the cinnamic acid residue, the quinoline ring and the lipophilic regions of the active site. The molecule of salubrinal effectively interacts with the GADD34:PP1 enzyme closing access to the active site. The energy of the complex GADD34:PP1-salubrinal forms - 12.2489 kcal/mol. The salubrinal molecule is additionally fixed in the active center of the enzyme due to the formation of an intermolecular hydrogen bond involving the amino acid Tyr 272 (Fig. 4b). A hydrogen bond arises between the nitrogen atom of the pyridine type of the quinoline ring and the hydroxyl group of Tyr 272 (the length of the N...HO bond is 3.432 Å). The salubrinal molecule is also fixed due to the shortened intermolecular polar contacts: 1) between the oxygen atom of the amide fragment and the hydroxyl group Tyr 134; 2) between the sulfur atom of the thiourea fragment and the guanidine fragment Arg 221.



**Figure 4.** a) The structure of the GADD34:PP1 holoenzyme. Protein phosphatase 1 (PP1) is represented in white color, and GADD34 - pink light. In the active site of the holoenzyme, there is a phosphoric acid residue, depicted in the form of spheres; b) the orientation of the *salubrinal* molecule in the active site of the GADD34:PP1 holoenzyme according to molecular docking data.

To determine the effect of the residue of cinnamic acid and quinoline cycle in the salubrinal preparation on the ability to bind to the active site of the GADD34:PP1 enzyme, we periodically replaced one of the fragments with other groups. According to the molecular docking data, six compounds out of fifty-four analyzed salubrinal analogues (*see supporting information Table S3 and S4*) exceeded it in the stability of the complex formed with GADD34:PP1 (Fig. 5).

The most stable complex with GADD34:PP1 is formed by (*E*)-3-(thiophen-2-yl)-*N*-(2,2,2-trichloro-1-(3-(quinolin-8-yl)thioureido)ethyl)acrylamide (**S1**) (Fig. 6a), the energy of the complex with GADD34:PP1 is -12.8833 kcal/mol, RMSD 1.4 Å. The molecule of the compound (**S1**) is additionally fixed in the enzyme active site due to the intermolecular hydrogen bond formed between the nitrogen atom of the thiourea fragment and the -OH group of Tyr 272, the length of the HN...HO bond is 3.605 Å). It is also fixed due to the formation of shortened intermolecular polar contacts: 1) between the oxygen atom of the amide fragment and the hydroxyl group of Tyr 134; 2) between the sulfur atom of the thiourea fragment and the guanidine fragment Arg 221.

The energy of the complex *N*-(2,2,2-trichloro-1-(3-(2-chlorophenyl)thioureido)ethyl)cinnamamide (**S2**) with GADD34:PP1 forms -12.5738 kcal/mol, RMSD 2.3 Å (Fig. 6b). The molecule of the compound (**S2**) is additionally fixed in the active site of the enzyme due to the formation of two intermolecular hydrogen bonds involving amino acids His 125 and Asn 124 (Fig. 6b). Both hydrogen bonds are formed by the oxygen atom of the amide fragment. In the first case, the hydrogen bond is with the pyrrole atom of nitrogen of the imidazole ring His 125 (the C=O...HN bond length is 2.133 Å), and in the second case - with the amide fragment of amino acid Asn 125 (the C=O...H<sub>2</sub>NC(O) bond length is 2.884 Å). The molecule of the compound (**S2**) is also fixed due to the shortened intermolecular polar contact between the sulfur atom of the thiourea fragment and the guanidine fragment Arg 221.



**Figure 5.** Structures of salubrinal analogues, surpassing it in the strength of the salubrinal preparation superior to the strength of the formed complex with the GADD34:PP1 holoenzyme.

*N*-(2,2,2-Trichloro-1-(3-(naphthalen-2-yl)thioureido)ethyl)cinnamamide (**S3**) forms the complex with the GADD34:PP1 enzyme having the energy of -12.4218 kcal/mol, RMSD 1.4 Å (Fig. 6c). The molecule of the compound (**S3**) is additionally stabilized in the enzyme active center due to the intermolecular hydrogen bond formed by the oxygen atom of the oxygen amide fragment and the HN group of Tyr Arg 221, the C=O...HN bond length is 2.481 Å. The molecule of the compound (**S3**) is also stabilized due to the shortened intermolecular polar contact between the sulfur atom of the thiourea fragment and the guanidine fragment Arg 221.

The energy of the complex *N*-(2,2,2-trichloro-1-(3(naphthalene-1-yl)thioureido)ethyl)cinnamamide (**S4**) with GADD34:PP1 forms -12.4195 kcal/mol, RMSD 4.8 Å (Fig. 6d). The molecule of the compound (**S4**) is additionally stabilized in the enzyme active center due to the formation of intermolecular hydrogen bonds: 1) between the oxygen atom of the amide fragment and the -OH group of Tyr 134, the C=O...HO bond length is 2.758 Å; 2) between the nitrogen atom of the thiourea fragment and the -OH group of Tyr 272, the HN...HO bond length is 2.999 Å. Moreover, stabilization occurs due to the shortened intermolecular polar contact between the sulfur atom of the thiourea fragment and the guanidine fragment Arg 221.

(*E*)-3-(Thiophen-3-yl)-*N*-(2,2,2-trichloro-1-(3-(quinolin-8-yl)thioureido)ethyl)acrylamide **(S5)** and *N*-(2,2,2-trichloro-1-(3-(4-chlorophenyl)thioureido)ethyl)cinnamamide **(S6)** form complexes with the GADD34:PP1 enzyme having the energy of -12.3286 kcal/mol (RMSD - 1.3 Å) and -12.3140 kcal/mol (RMSD - 5.4 Å), respectively (Fig. 6e, 6f). The compounds **(S5)** and **(S6)** do not form intermolecular hydrogen bonds in the active center of GADD34:PP1, their interaction with amino acids forming the active site is obviously hydrophobic in nature. In this case, the formation of weak shortened intermolecular polar contacts is possible, for example, between the sulfur of the thiourea fragment and the guanidine fragment Arg 221. It should be noted, that the molecule of the compound **(S6)** in the active center of GADD34:PP1 is rotated 180° as compared to the salubrinal molecule and the remaining compounds hits.

Figure 7 shows that for both quinoline derivatives and cinnamic acid derivatives, the energy of the complex formed is clearly related to GADD34:PP1 from the RMSD value. The quinoline derivatives interact closely with the active site of the enzyme, the RMSD value does not exceed 3.5 Å. While for the cinnamic acid derivatives, RMSD can vary from 1.5 to 12.0 Å. This is due to the fact that there are

two lipophilic sites in the active center of the enzyme. The quinoline cycle, due to spatial difficulties, is clearly fixed only in one of them, and the cinnamic acid residue can interact with both. This can lead to a reversal of the inhibitor molecule located in the active center by 180° relative to the salubrinal molecule, which is observed, for example, for the compound (**S6**).



**Figure 6.** Position of the molecules of the compounds **(S1)**-**(S6)** in the active center of the GADD34:PP1 holoenzyme.



Figure 7. Energy dependence of the complex GADD34:PP1-Inhibitor on the RMSD value. Color: a) ■ quinoline derivatives; ■ cinnamic acid derivatives; ■ salubrinal, taken into account when constructing a linear regression line in both cases.

Based on our findings, when searching for the GADD34:PP1 inhibitors, other than cinnamic acid and quinoline derivatives, special attention should be paid to the compounds containing a naphthalene and isoquinoline ring, heterocyclic analogs of cinnamic acid, and compounds containing chlorine atoms in the aromatic ring. The results of our work are in good agreement with the already published experimental data on establishing the dependence of the structure-activity of salubrinal analogues [39,54]. For example, the low activity of 2-amino-pyridine derivatives, for which the EC<sub>50</sub> lies in the range of 28-72  $\mu$ M [39], as compared to the derivatives of 8-aminoquinoline (EC<sub>50</sub> = 15-16  $\mu$ M) [39,54], can be explained by the high energy of the complex that they form with GADD34:PP1. The lower energy of the GADD34:PP1-Inhibitor complex can also explain the high efficiency of (*E*)-3-(thiophen-2-yl)acrylamide derivatives (EC<sub>50</sub> = 4-43  $\mu$ M) compared with cinnamamide derivatives (EC<sub>50</sub> = 6-57  $\mu$ M) [54].

## Conclusions

In this paper, the search for new analogues of salubrinal has been carried out by molecular modeling. We have shown that the most stable conformation of the salubrinal molecule and its analogues contains the intramolecular hydrogen bond between the hydrogen atom of the thiourea fragment and the oxygen atom of the amide fragment. The binding site of salubrinal to the active site of the enzyme has been established. We have found the compounds, which form stronger complexes with the enzyme than salubrinal itself. The results of this work can be used to create new phosphatase inhibitors of the eukaryotic translation initiation factor GADD34:PP1.

Conflict of interest: The authors declare no conflicts of interests.

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**Supporting Information for the paper:** Molecular docking studies of salubrinal and its analogs as inhibitors of the GADD34:PP1 enzyme

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Table S1. The results of geometry optimization of salubrinal analogues containing cinnamic acid residue

 $CCl_3 S$ N R 

Entry	R	CAS No	E, kcal/mol	The length of the NHO=C bond, Å	Angle value C=OH, degrees	Time, s
1		405060-59-9	-111285.9025	1.891	108.33	347
2		863036-22-0	-111285.2829	1.880	108.88	348
3		294654-78-7	-110631.9704	2.273	105.11	344
4		324769-18-8	-110634.7191	1.863	108.63	365
5	NH <sub>2</sub>	405060-99-3	-98255.3837	1.919	109.47	192

6	CH3	405060-96-0	-104370.9745	1.883	108.96	207
7	$\begin{array}{c} \downarrow \\ \downarrow \\ \downarrow \\ \downarrow \\ \downarrow \\ \downarrow \\ O \\ CH_3 \\ H_3 \end{array} $	863036-35-5	-129157.9187	1.943	108.52	736
8		301359-85-3	-99001.5764	1.891	108.31	232
9	$\left  \begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\$	301359-95-5	-102453.0936	1.863	108.48	272
10		301359-86-4	-102454.5158	1.864	108.24	240
11		301359-87-5	-102454.6059	1.868	108.08	249
12	$H_{3}C$ $CH_{3}$	1429483-71-5	-105904.3082	1.863	108.55	315
13	$ CH_3$ H <sub>3</sub> C	301359-93-3	-105905.3171	1.865	108.40	288
14	$\langle - \rangle $ $H_3$ $H_3$ $C$	301359-94-4	-105905.2465	1.861	108.55	287

15	HO	294657-79-7	-105777.6924	1.858	107.98	247
16	- OH	3037775-31-7	-105778.1081	1.890	108.48	271
17		405060-94-8	-103291.5220	1.863	108.80	246
18	$H_3CO$	301359-88-6	-109214.6095	1.874	108.24	288
19	$\rightarrow$ $         -$	1346508-38-0	-109215.6380	1.886	108.42	298
20	HOOC	303775-35-5	-115287.0750	1.858	108.47	295
21	СООН	294655-14-4	-115289.4152	1.870	108.75	318
22	$\sim$	294655-12-2	-118722.8463	1.858	108.12	359
23		294654-81-2	-115870.4210	1.871	108.65	291

24		294654-82-3	-115870.3348	1.868	108.55	301
25		301359-89-7	-105946.6228	1.864	108.36	241
26		301359-90-0	-105947.3592	1.860	108.50	224
27		301359-91-1	-105947.5373	1.859	108.49	234
28	Br	301359-92-2	-106797.8334	1.860	108.54	237
29		301359-97-7	-112891.5613	1.862	108.63	254
30		301359-98-8	-112891.2835	1.863	108.60	277
31	$\langle - \rangle $	301359-96-6	-138810.5173	1.859	108.45	355
32	$\sim \sim $	301815-13-4	-116057.8543	1.863	108.53	375

33		294653-17-1	-124258.2307	1.896	108.75	723
34	$\sim \sim $	294654-77-6	-120926.7177	1.862	108.67	409
35		863036-23-1	-99655.3253	1.857	108.48	215
36	$\sim \sim $	1349267-41-9	-110899.9614	1.873	108.85	294
37		1349267-41-9	-112093.4153	2.339	104.89	272
38		1346508-37-9	-97718.9301	2.507	105.22	233
39		1346508-36-8	-101035.0119	2.521	105.03	225
40	∠N CH <sub>3</sub>	1429483-72-6	-101809.4188	2.493	105.51	275

Table S2. The results of geometry optimization of salubrinal analogues containing quinoline ring



	P			The length of the	Angle value	Time,
Entry	ĸ	CAS NO	E, KCal/mol	NHO=C bond, Å	C=OH, degrees	seconds
1	CH <sub>3</sub> -	294658-37-0	-90772.1333	1.835	109.78	141
2	$H_{3}C$ $H_{3}C$	324769-75-5	-97666.2121	1.849	109.08	207
3	H <sub>3</sub> C CH <sub>3</sub>	294658-28-9	-101115.4287	1.832	109.23	261
4	(CH <sub>3</sub> ) <sub>3</sub> C-	412962-51-7	-101113.4907	1.843	109.50	247
5	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	305856-11-5	-108568.4619	2.370	101.87	319
6		294646-80-3	-105112.9884	2.387	101.67	291
7	$\langle \overline{\rangle} \rangle$ CH <sub>3</sub>	324017-95-0	-108568.5668	2.460	101.65	323
8	$H_{3}C$	330684-99-6	-108574.0942	2.261	105.45	330

9	$H_3C$	294658-30-3	-108574.2167	2.418	105.03	313
10	$\operatorname{Br}$	330567-60-7	-112916.8710	2.288	104.53	308
11	$H_3CO \longrightarrow $	294658-45-0	-115333.8079	2.382	102.98	363
12		405060-98-2	-116754.1849	2.262	104.94	437
13	C C C C C C C C C C C C C C C C C C C	324017-57-4	-105701.1794	2.347	102.79	246
14	S	1346508-21-1	-109395.6421	1.901	108.14	336
15	s s s	1346508-22-2	-109398.3840	1.913	107.96	313

Table S3. The results of molecular docking of salubrinal analogues containing cinnamic acid residue



Entry	P		E kcal/mol		Time,
Entry	n.	CAS NO	E, KCdi/1101	RIVISD, A	seconds
1		405060-59-9	-12.2489	-	68
2		863036-22-0	-11.4506	9.8	64
3		294654-78-7	-12.4195	4.8	66
4		324769-18-8	-12.4218	1.4	68
5	V NH2	405060-99-3	-10.4440	8.6	222

6	CH3	405060-96-0	-10.9403	8.7	712
7	$\begin{array}{c} & H \\ & N \\ & & O \\ & & CH_3 \\ & O \\ & CH_3 \end{array}$	863036-35-5	-11.0821	6.3	25055
8		301359-85-3	-11.0494	2.4	91
9	$H_{3C}$	301359-95-5	-11.8415	2.5	78
10		301359-86-4	-11.7071	3.4	81
11		301359-87-5	-10.9713	9.5	82
12	$H_3C$ CH <sub>3</sub>	1429483-71-5	-11.4002	3.5	70
13	$ CH_3$ H <sub>3</sub> C	301359-93-3	-11.0643	2.6	74
14	$\begin{array}{c} \begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\$	301359-94-4	-10.7865	6.86	70

15	HO	294657-79-7	-11.9634	2.8	76
16	->>ОН	3037775-31-7	-10.8881	6.7	80
17	HS	405060-94-8	-11.2845	2.0	75
18	$H_3CO$	301359-88-6	-11.3522	3.6	152
19		1346508-38-0	-9.8336	5.5	182
20	HOOC	303775-35-5	-11.7954	3.4	176
21	СООН	294655-14-4	-10.0874	3.4	202
22	O O OCH <sub>3</sub>	294655-12-2	-11.1046	3.7	501
23		294654-81-2	-9.9688	4.2	201

24		294654-82-3	-10.2628	11.0	195
25		301359-89-7	-12.5738	2.3	86
26		301359-90-0	-10.2843	3.2	88
27	->-Cl	301359-91-1	-12.3140	5.4	88
28	Br	301359-92-2	-10.4781	10.6	87
29		301359-97-7	-10.5242	9.6	82
30		301359-98-8	-11.0901	9.8	77
31	$\langle - \rangle $ $CF_3$ $CF_3$	301359-96-6	-10.5026	8.7	140
32	$\sim \sim $	301815-13-4	-10.9633	6.9	178

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33		294653-17-1	-11.6777	6.8	45
34	$\sim \sim \sim \sim \sim \sim \circ \circ$	294654-77-6	-10.2446	10.0	108
35		863036-23-1	-9.8478	11.1	87
36	Br CH <sub>3</sub>	1349267-41-9	-10.0146	11.9	60
37		1349267-41-9	-9.5065	6.9	60
38		1346508-37-9	-10.1080	4.1	35
39		1346508-36-8	-10.0909	6.1	35
40	$\sum_{N}^{N} N^{CH_3}$	1429483-72-6	-10.3209	5.5	31

Table S4. The results of molecular docking of salubrinal analogues containing quinoline ring

CCl<sub>3</sub> S  $\mathbf{O}$ N H `N´ H Н

Entry	R	CAS No	E, kcal/mol	RMSD, Å	Time,
					seconds
1	CH <sub>3</sub> -	294658-37-0	-9.5079	3.4	30
2	$H_3C$ $H_3C$	324769-75-5	-9.5363	3.3	37
3	H <sub>3</sub> C CH <sub>3</sub>	294658-28-9	-10.1110	3.4	48
4	(CH₃)₃C-	412962-51-7	-10.5443	2.4	35
5	<u></u> .	305856-11-5	-11.0971	3.5	55
6		294646-80-3	-12.0377	1.3	38
7	$\langle \qquad \rangle$ CH <sub>3</sub>	324017-95-0	-12.0111	3.3	34
8	$H_{3}C$	330684-99-6	-10.2993	2.6	37

9	$H_3C$	294658-30-3	-10.9900	1.9	36
10	$\operatorname{Br}$	330567-60-7	-11.9627	3.4	36
11	$H_3CO \longrightarrow $	294658-45-0	-10.4252	2.4	54
12		405060-98-2	-11.1993	2.4	35
13		324017-57-4	-9.8686	1.4	38
14	S	1346508-21-1	-12.8833	1.4	74
15		1346508-22-2	-12.3286	1.3	75